

ME Capstone

Case Studies & Insights into Sustainable Design

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Feb 19, 2024

Outline:

Case Studies and Insights into Sustainable Design

- ▶ UN SDGs
- ▶ A Cup of Tea?
- ▶ Comparing Unlike Energy Sources?
- ▶ Are EVs Emissions Free?

SUSTAINABLE DEVELOPMENT GOALS



UN Sustainable Development Goals

UN SDGs

- ▶ The Sustainable Development Goals (SDGs), were adopted by the United Nations in 2015 as a **universal call to action** to end poverty, protect the planet, and ensure that by 2030 all people enjoy peace and prosperity.
- ▶ The 17 SDGs are **integrated**—they recognize that action in one area will affect outcomes in others, and that development must balance social, economic and environmental sustainability.
- ▶ Countries have committed to **prioritize progress** for those who're furthest behind. The SDGs are designed to end poverty, hunger, AIDS, and discrimination against women and girls.
- ▶ The **creativity, knowhow, technology and financial resources** from all of society is necessary to achieve the SDGs in every context.

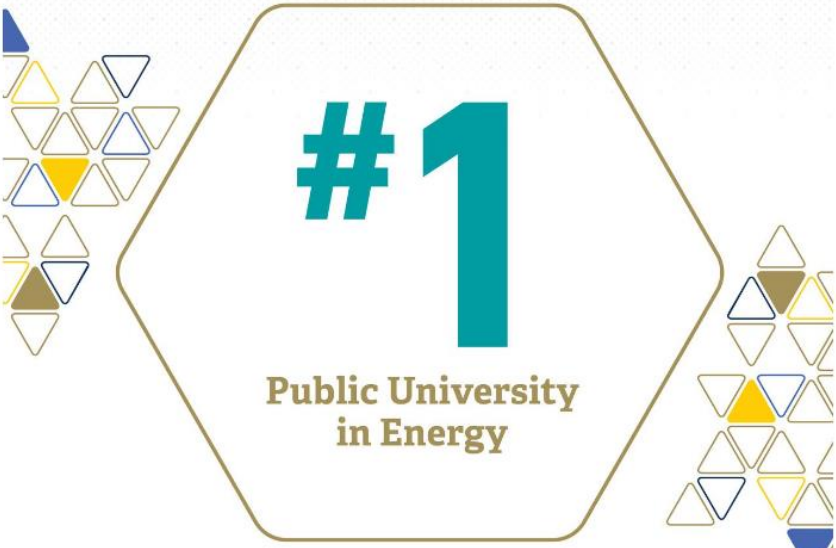

▶ Trending News...

anked-public-university-energy

[Home](#) > Georgia Tech Named Top-Ranked Public University in Energy

Georgia Tech Named Top-Ranked Public University in Energy

FEB 07, 2024 -- ATLANTA, GA



#1
Public University
in Energy

U.S. News and World Report

U.S. News & World Report has **ranked** the Georgia Institute of Technology as the top public university and No. 3 nationally in energy and fuels research. This is the first year the category has been included in the annual rankings, and Georgia Tech's dominance reflects the dynamic research and expertise of the Institute.

"I'm thrilled to see Georgia Tech recognized for our leading-edge approach to creating sustainable energy solutions," said Executive Vice President for Research Chaesiki Abdallah. "This achievement reflects the unwavering commitment of our faculty and researchers to conducting groundbreaking research."



<https://research.gatech.edu/georgia-tech-named-top-ranked-public-university-energy>



also,

- ▶ GT Sustainability Showcase during SDG Week...

Sustainability Showcase

SDG Week | March 4 - 8, 2024



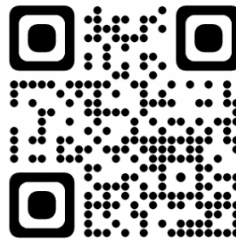
Georgia Tech
Brook Byers Institute
for Sustainable Systems



Office of
Sustainability

- ◆ Lightning Talks
- ◆ Panel Discussions
- ◆ Posters
- ◆ Art on Display

For details and updates, visit:
sustainable.gatech.edu/showcase



SUSTAINABLE
DEVELOPMENT GOALS

Learn about all
the SDG Week
events at:

sustain.gatech.edu/sdg-week



March 6-7, 8:30-4pm, in the John Lewis Student Center's Atlantic Theater.



Poll Question 1

- ▶ Suggest a method for boiling a cup of water

Some Methods for Boiling Water



Microwave



Gas Stovetop



Electric Kettle



Concentrated Solar*

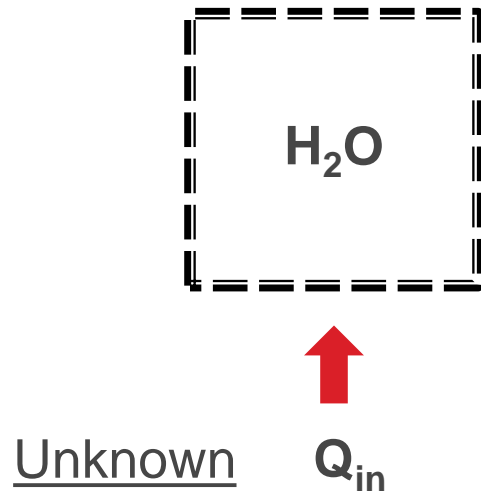


Wood Fire
(Open Flame)

*Source: Contemporary Energy Solar Kettle & Amazon

Poll Question 2

- ▶ Suggest an equation for estimating the energy required to boil a cup of water



$$Q_{in} = m \cdot c_p \cdot \Delta T$$

Knowns and Assumptions

Substance = Water

$m = 8.0$ fl. oz. (1 cup)

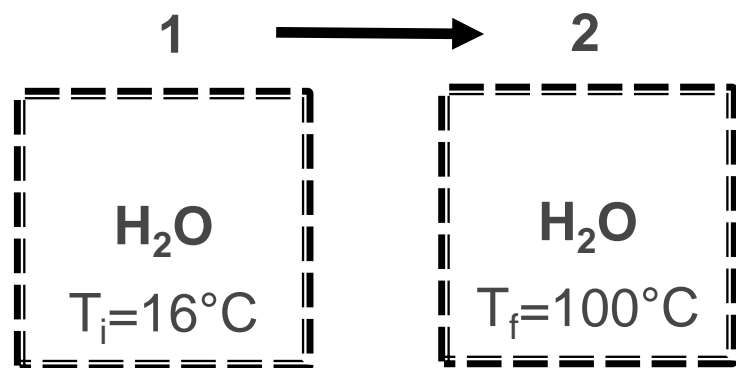
$m = 0.227$ kg

$c_p = 4.184$ kJ/kg $^{\circ}$ C

$T_i = 16^{\circ}$ C

$T_f = 100^{\circ}$ C

Theoretical Energy Input



$$Q_{in} = m \cdot c_p \cdot \Delta T$$

$$Q_{in} = (0.227\text{kg}) \cdot (4.184\text{kJ/kg}^\circ\text{C}) \cdot (100 - 16^\circ\text{C})$$

$$Q_{in} = 79.7 \text{ kJ}$$

Conversion: 1 kJ = 0.2777 Wh

$$Q_{in} = 22.1 \text{ Wh}$$

How much energy is this?

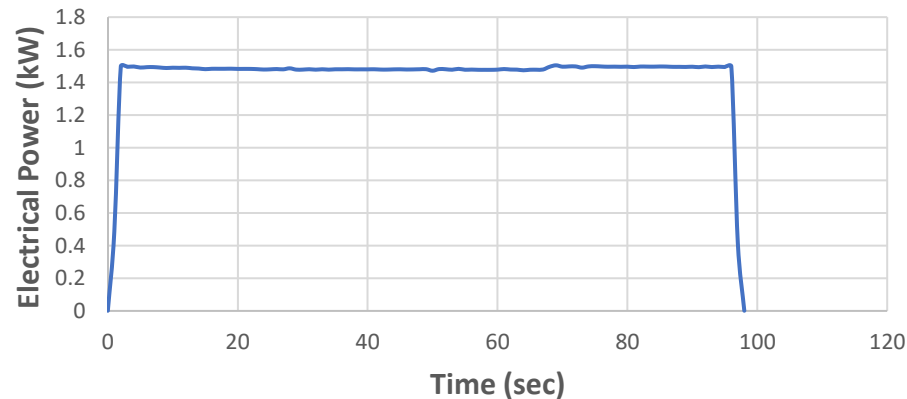
REF: $W_{elec} = 10 \text{ W}\cdot\text{h}$ (to charge a smart phone)

REF: $W_{elec} = 37,000 \text{ W}\cdot\text{h}$ (to charge a Nissan Leaf)

Boiling Water in an Electric Tea Kettle

Tea Kettle Experiment

Instantaneous Power to Boil 8oz of Water
in Electric Tea Kettle



$$W_{\text{elec}} \approx 39.4 \text{ Wh}$$

$$Q_{\text{min, theor}} = 22.1 \text{ Wh}$$

$$\eta = E_{\text{min}}/E_{\text{actual}}$$

$$\eta_{\text{elec_kettle}} \approx 39.4\text{Wh}/57.5\text{Wh}$$

$$\eta_{\text{elec_kettle}} \approx 0.56 \text{ (56\%)}$$

$$W_{\text{elec}} = P \cdot \Delta t$$

$$W_{\text{elec}} \approx (1500 \text{ W}) \cdot (96 \text{ sec})$$

$$W_{\text{elec}} \approx 142,000 \text{ W} \cdot \text{s}$$

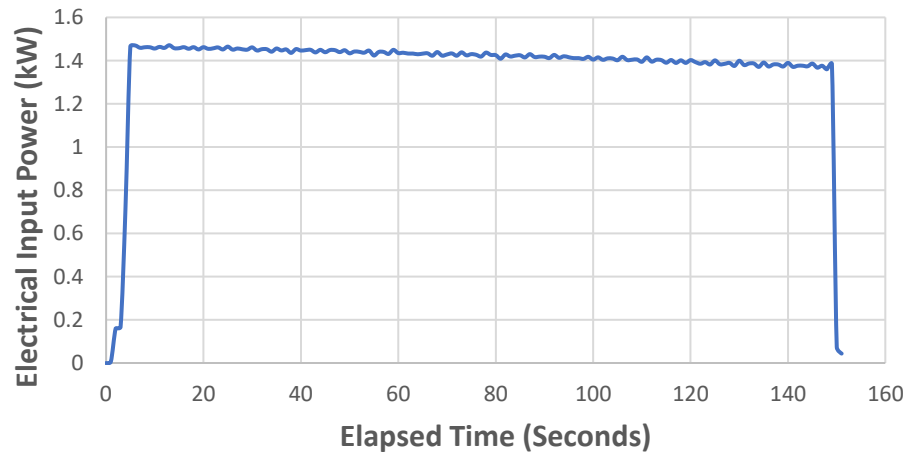
$$W_{\text{elec}} \approx 39.4 \text{ W} \cdot \text{h}$$

Boiling Water in a Microwave

Microwave Experiment



Energy to Boil 8 oz Water in a Microwave



$$W_{\text{elec}} \approx 57.5 \text{ Wh}$$

$$Q_{\text{min, theor}} = 22.1 \text{ Wh}$$

$$\eta = E_{\text{min}} / E_{\text{actual}}$$

$$\eta_{\text{microwave}} \approx 22.1 \text{ Wh} / 57.5 \text{ Wh}$$

$$\eta_{\text{microwave}} \approx 0.38 \text{ (38\%)}$$

$$W_{\text{elec}} = P \cdot \Delta t$$

$$W_{\text{elec}} \approx (1380 \text{ W}) \cdot (150 \text{ sec})$$

$$W_{\text{elec}} \approx 207,000 \text{ W} \cdot \text{s}$$

$$W_{\text{elec}} \approx 57.5 \text{ W} \cdot \text{h}$$

Boiling Water on a Gas Burner



Gas Burner Experiments



Open Pot

5000 BTU/hr (1466 W)

Time to boil \approx 390 s (0.108h)

$$W_{\text{gas,low,open}} = \dot{Q}_{\text{in}} \cdot \Delta t$$

$$W_{\text{gas,low,open}} \approx (1466 \text{ W}) \cdot (390\text{s})$$

$$W_{\text{gas,low,open}} \approx 572,000 \text{ W} \cdot \text{s}$$

$$W_{\text{gas,low,open}} \approx 159 \text{ W}\cdot\text{h}$$

$$\eta_{\text{gas,low,open}} \approx 0.14 \text{ (14\%)}$$

Closed Pot

9500 BTU/hr (2786 W)

Time to boil \approx 120 s (0.033h)

$$W_{\text{gas,high,closed}} = \dot{Q}_{\text{in}} \cdot \Delta t$$

$$W_{\text{gas,high,closed}} \approx (2786 \text{ W}) \cdot (120\text{s})$$

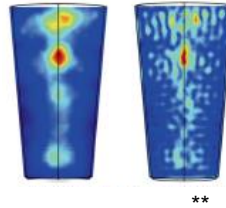
$$W_{\text{gas,high,closed}} \approx 334,000 \text{ W} \cdot \text{s}$$

$$W_{\text{gas,high,closed}} \approx 93 \text{ W}\cdot\text{h}$$

$$\eta_{\text{gas,high,closed}} \approx 0.24 \text{ (24\%)}$$

Summary

Electricity



$$W_{\text{elec}} \approx 57.5 \text{ W}\cdot\text{h}$$
$$\eta_{\text{microwave}} \approx 0.38 \text{ (38\%)}$$



$$W_{\text{elec}} \approx 39.4 \text{ W}\cdot\text{h}$$
$$\eta_{\text{elec_kettle}} \approx 0.56 \text{ (56\%)}$$

Natural Gas



$$W_{\text{gas,low,open}} \approx 159 \text{ W}\cdot\text{h}$$
$$\eta_{\text{gas,low,open}} \approx 0.14 \text{ (14\%)}$$

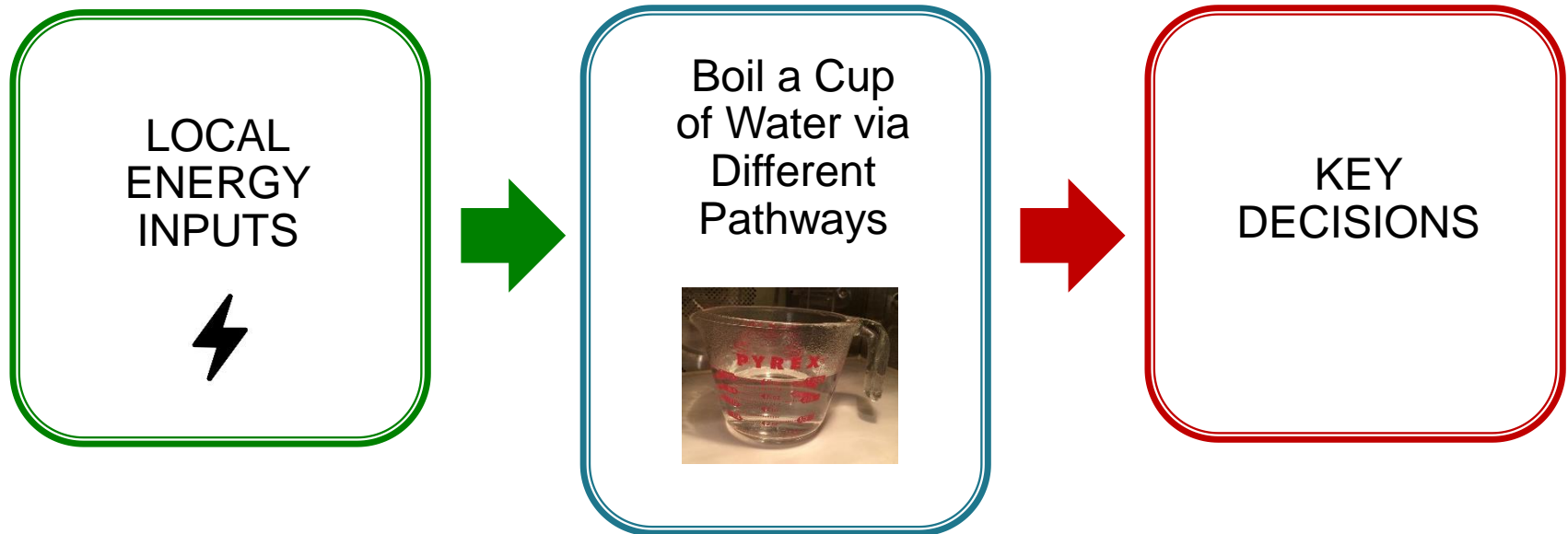


$$W_{\text{gas,high,closed}} \approx 93 \text{ W}\cdot\text{h}$$
$$\eta_{\text{gas,high,closed}} \approx 0.24 \text{ (24\%)}$$

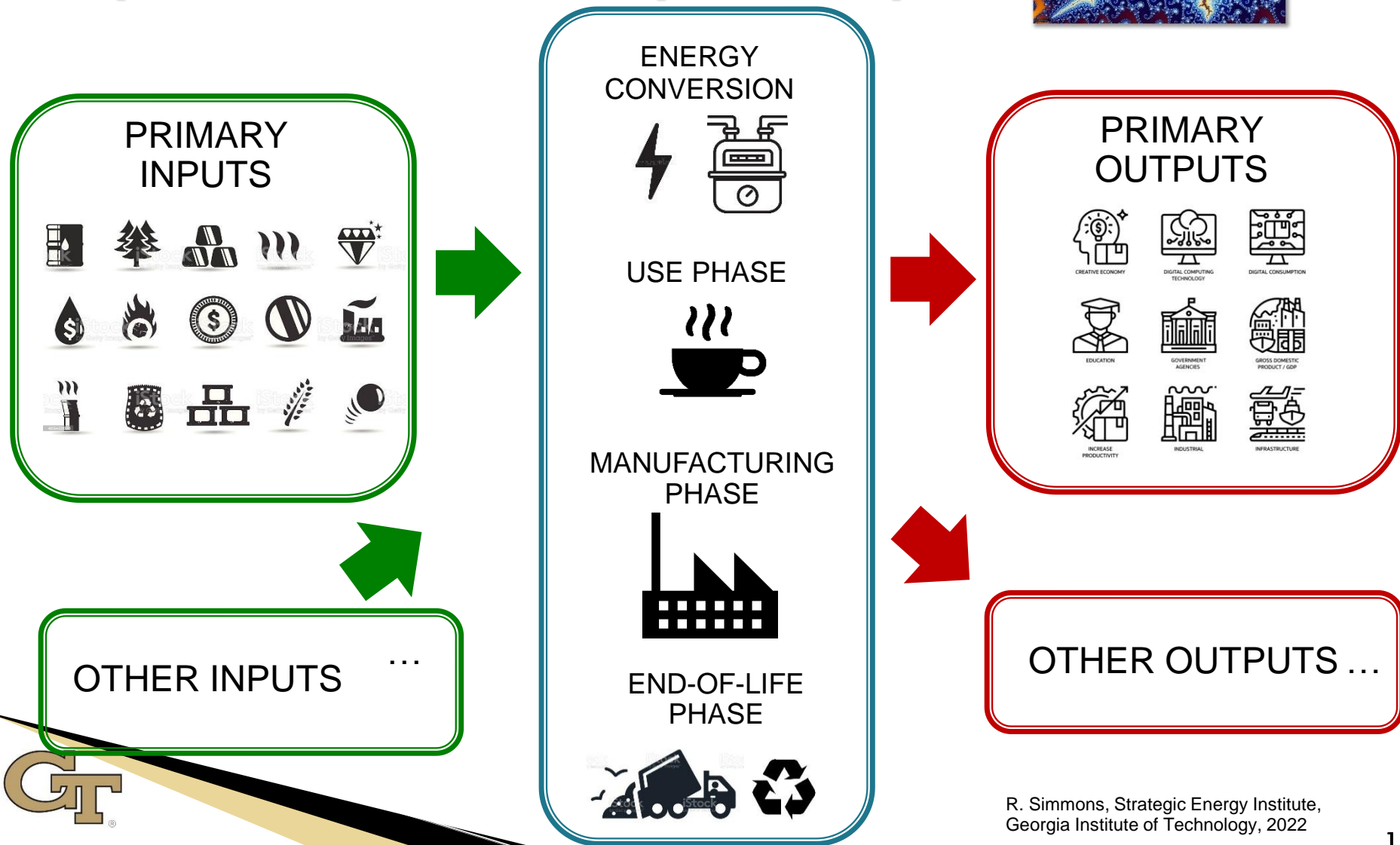
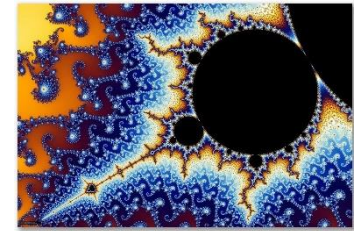
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Nature Reviews Physics | <https://doi.org/10.1038/s42254-020-0233-1> | Published online 13 August 2020

Engineering 101: System Boundary is Key



Engineering 101: System Boundary is Key



Consider the Boundary & Upstream Impacts



$$W_{\text{elec}} \approx 39.4 \text{ W}\cdot\text{h}$$

$$\eta_{\text{elec_kettle}} \approx 0.56 \text{ (56\%)}$$

Electricity_home

Electricity_Distribution

$$\eta_d \approx 0.98 \text{ (98\%)}$$

Electricity_Transmission

$$\eta_T \approx 0.94 \text{ (94\%)}$$

Electricity_Generation

$$\eta_G \approx 0.10 \text{ to } 0.95 \text{ (~40\%)}$$

$$\eta_{\text{elec_kettle,net,sys}} \approx \eta_{\text{elec_kettle}} \eta_d \eta_T \eta_G \approx 0.20 \text{ (20\%)}$$

Consider the Boundary & Upstream Impacts



$W_{\text{gas,high,closed}} \approx 93 \text{ W}\cdot\text{h}$

$\eta_{\text{gas,high,closed}} \approx 0.24 \text{ (24\%)}$

NaturalGas_home

**NG_Distribution $\eta_d \approx$
0.99 (99%)**

**NG_Refining
 $\eta_R \approx 0.93 \text{ (~93\%)}$**

$\eta_{\text{gas,high,closed,net,sys}} \approx \eta_{\text{gas,high,closed}} \eta_d \eta_R \approx 0.22 \text{ (22\%)}$

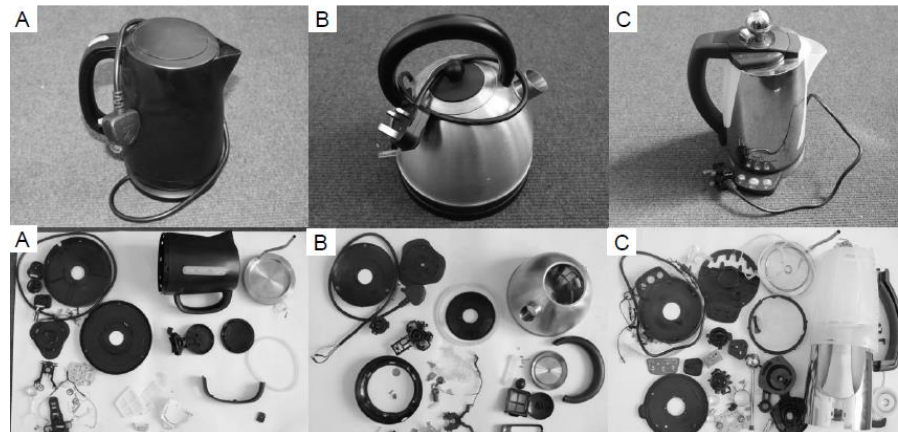
Poll Question 3

- ▶ What materials are used in an electric tea kettle?

Electric Kettle Raw Materials

Table 1. Inventory data for the plastic, metallic and eco-kettles

	Plastic kettle (polypropylene)	Metallic kettle (stainless steel)	Eco-kettle (polypropylene)	Data sources ^a
Raw materials				
Stainless steel (g)	248	645	342	Own measurement ^b
Brass (g)	27	24	26	~II~
Copper (g)	20	19	23	~II~
Aluminium (g)	-	-	34	~II~
Tin (g)	-	-	0.11	~II~
Silver (g)	-	-	0.02	~II~
Polypropylene (PP) (g)	467	419	841	~II~
Polyvinyl chloride (g)	58	57	58	~II~
Nylon (g)	66	36	29	~II~
Polyoxymethylene (POM) (g)	13	-	-	~II~
Polycarbonate (g)	9	-	75	~II~
Acrylonitrile butadiene styrene (g)	40	-	190	~II~
High density polyethylene (g)	-	-	7	~II~
Silicone (g)	16	1	37	~II~



Gallego Schmid, Alejandro, Life cycle environmental evaluation of kettles: Recommendations for the development of eco-design regulations in the European Union, The Science of the total environment, 2017

LCA – Approach

▶ Electric tea kettle LCA

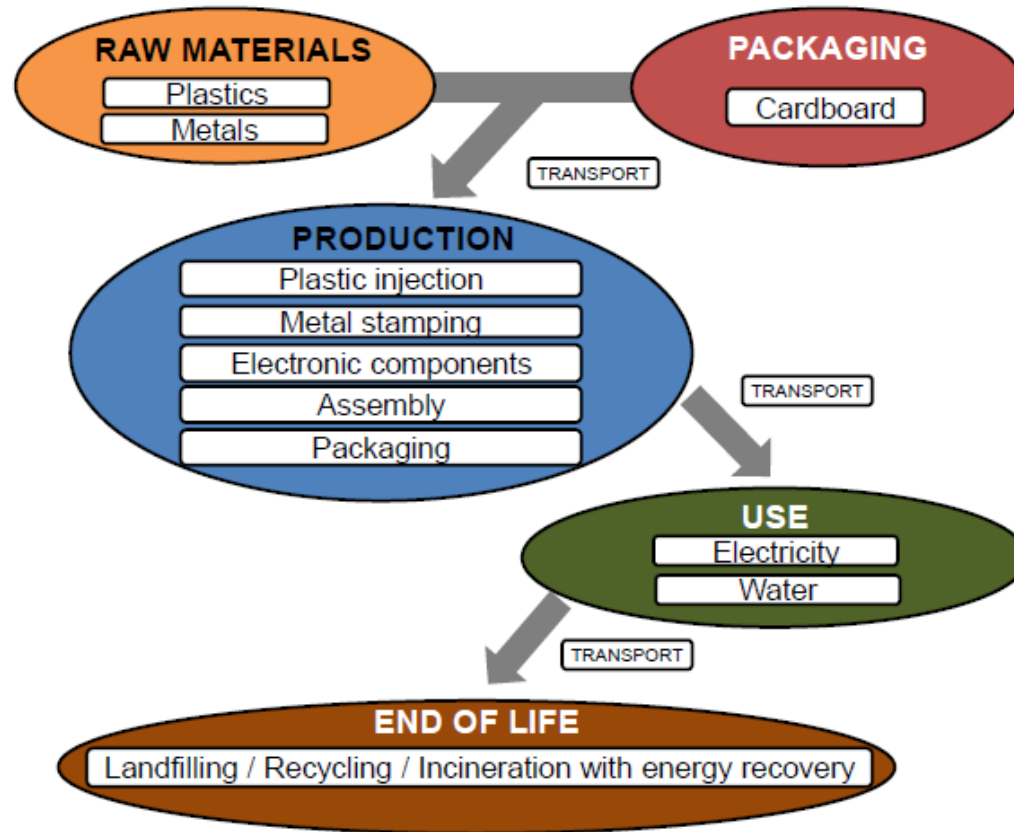
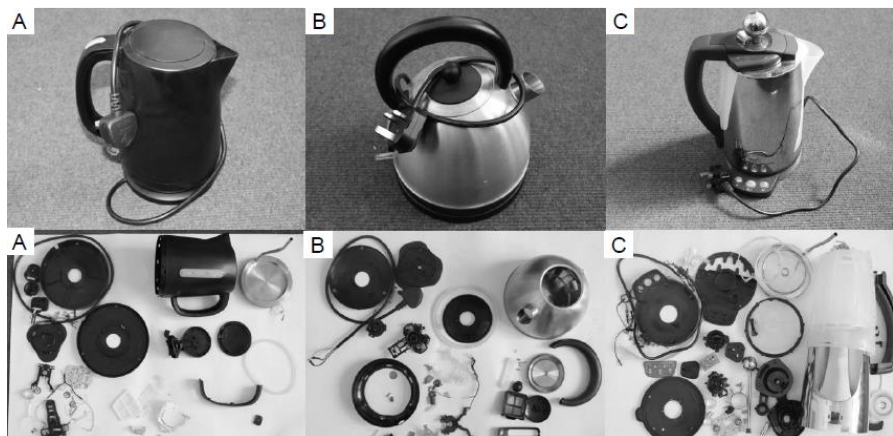


Fig. 1. System boundaries for the kettles considered in the study

Electric Kettle Energy/Process Inputs

Packaging				~II~
Low density polyethylene (g)	10	13	11	~II~
Cardboard boxes (g)	380	500	325	~II~
Production				
Injection moulding (heat) (MJ)	3.0	2.3	5.5	Manufacturer ^c
Injection moulding (electricity) (kWh)	1.0	0.8	1.9	~II~
Metal stamping (heat) (MJ)	0.01	0.03	0.02	~II~
Metal stamping (electricity) (kWh)	0.1	0.3	0.2	~II~
Power cord, plugs, thermal socket and wire cables				~II~
Heat (MJ)	0.04	0.04	0.04	~II~
Electricity (kWh)	0.1	0.1	0.2	~II~
Packaging				~II~
Heat (MJ)	2.1	2.5	1.3	~II~
Electricity (kWh)	0.2	0.2	0.1	~II~
Assembly				
Water (L)	14	14	14	Fischer et al., 2014



Gallego Schmid, Alejandro, Life cycle environmental evaluation of kettles: Recommendations for the development of eco-design regulations in the European Union, The Science of the total environment, 2017

Electric Kettle Dist/Use/E-o-L

Distribution				
Raw materials: to factory (tkm)	0.1	0.2	0.2	Own assumption ^d
Packaging: to factory (tkm)	0.1	0.1	0.1	~II~
Kettle: factory to Shanghai (tkm)	0.2	0.3	0.3	~II~
Kettle: Shanghai Rotterdam (tkm)	28.4	36.0	42.0	Sea distance (2017)
Kettle: Rotherham to Munich (tkm)	1.1	1.4	1.7	Via Michellin (2017)
Kettle: distribution centre (Munich) to retailer (tkm)	0.2	0.3	0.3	Own assumption ^d
End of life: to treatment facility (tkm)	0.1	0.2	0.2	Own assumption ^e
Use				
Electricity (kWh)	829	829	532	Own measurements ^{b,f}
Water (L)	1542	1542	1028	Fischer et al. (2014)
End of life				
Recycling (plastics) (g)	243	191	447	Eurostat (2016b)
Recycling (metals) (g)	281	654	404	Kemna et al. (2011)
Recycling (cardboard) (g)	322	424	275	Eurostat (2016b)
Incineration with energy recovery (plastics) (g)	205	161	377	~II~
Incineration with energy recovery (cardboard) (g)	31	41	27	~II~
Landfilling (plastics) (g)	230	174	424	~II~
Landfilling (metals) (g)	15	34	21	Kemna et al. (2011)
Landfilling (cardboard) (g)	25	36	23	Eurostat (2016b)

Electric Kettle LCA

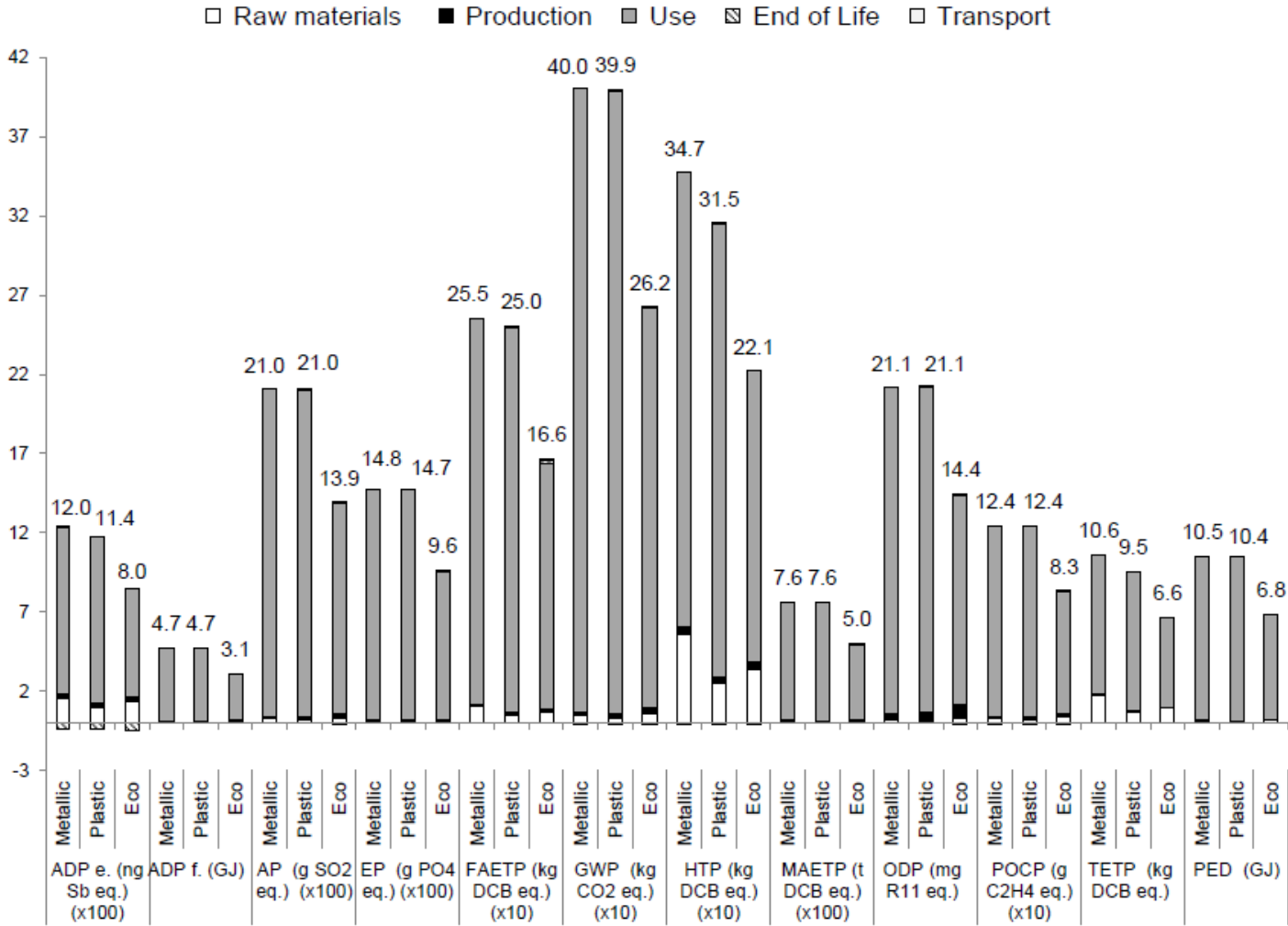
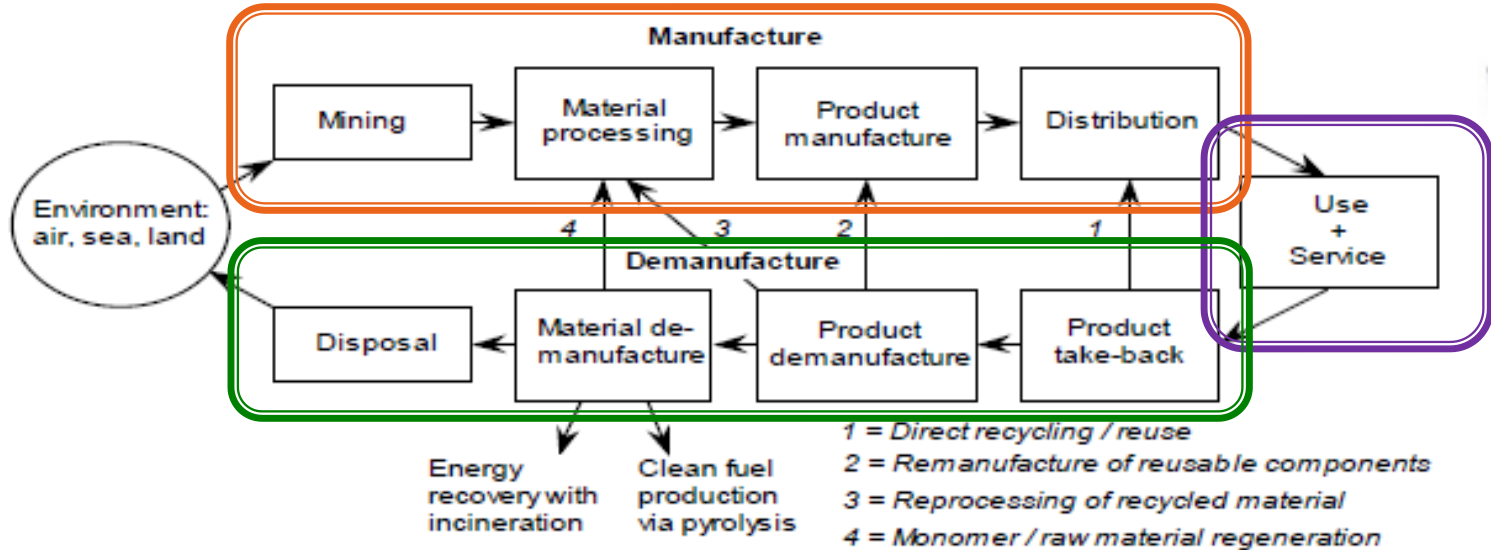


Fig. 3. Life cycle environmental impacts of plastic, metallic and eco-kettles over their average useful lifetime (4.4 years).

Poll Question 4

What share of total energy is typically used by a Plug-In EV during Use-Phase?

- ▶ Resource Extraction & Manufacturing (M)
- ▶ Use-Phase (U)
- ▶ End-of-Life Phase (EoL)



Flowchart source: Bras, B. "Sustainable Design and Manufacturing Including Environmental and Social Considerations," www.sdm.gatech.edu

LCA- Car Example

Typical breakdown for a plug-in hybrid EV

▶ Resource Extraction & Manufacturing

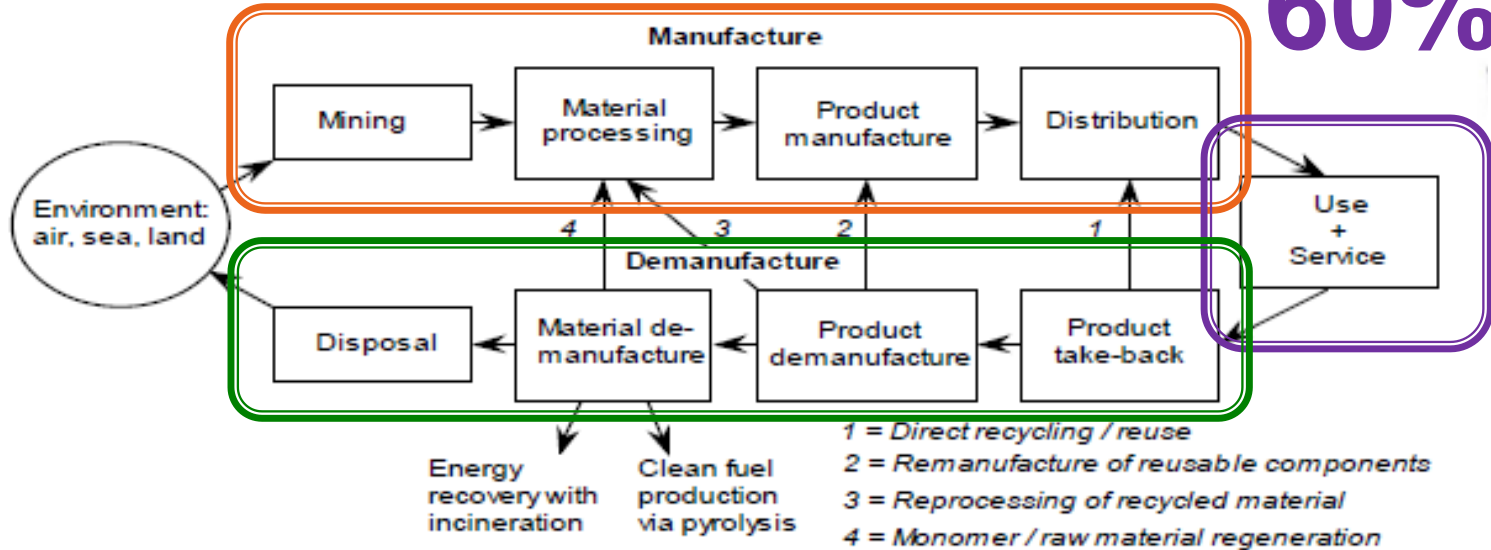
+

40%

▶ End-of-Life Phase

▶ Use-Phase

60%



Flowchart source: Bras, B. "Sustainable Design and Manufacturing Including Environmental and Social Considerations," www.sdm.gatech.edu

Compiled by: R. Simmons, Strategic Energy Institute, Georgia Institute of Technology, 2021

LCA- Car Example

Typical energy breakdown for a conventional ICE car

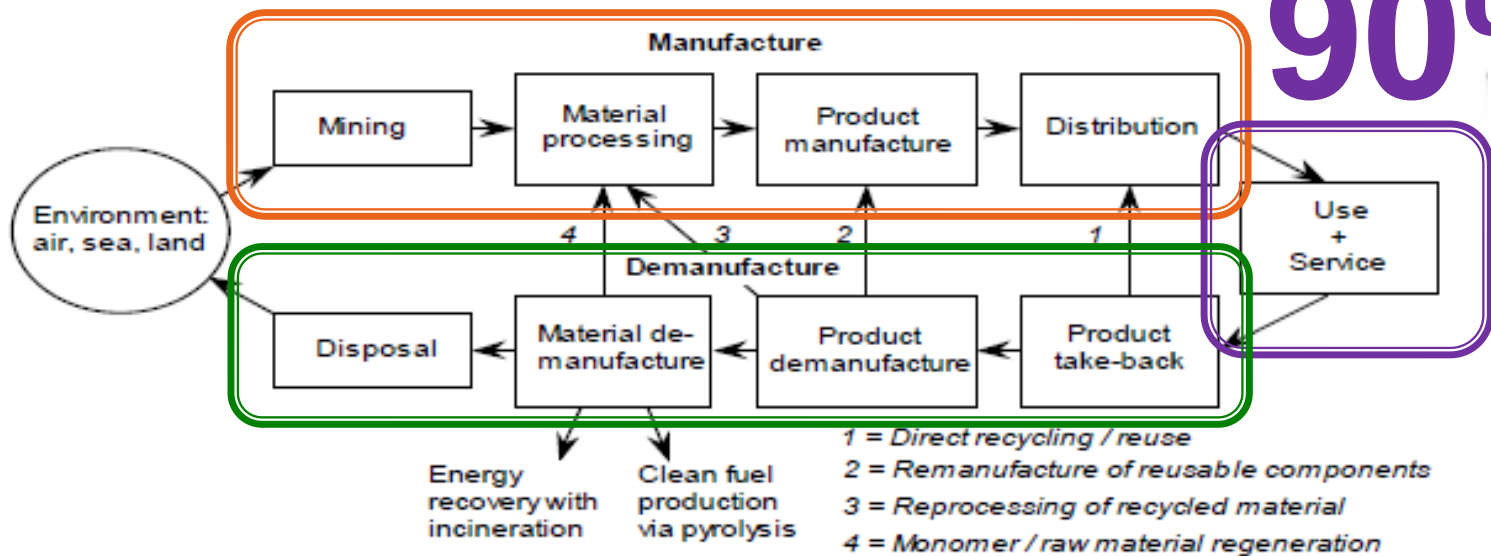
▶ Resource Extraction & Manufacturing

10%

▶ End-of-Life Phase

▶ Use-Phase

90%



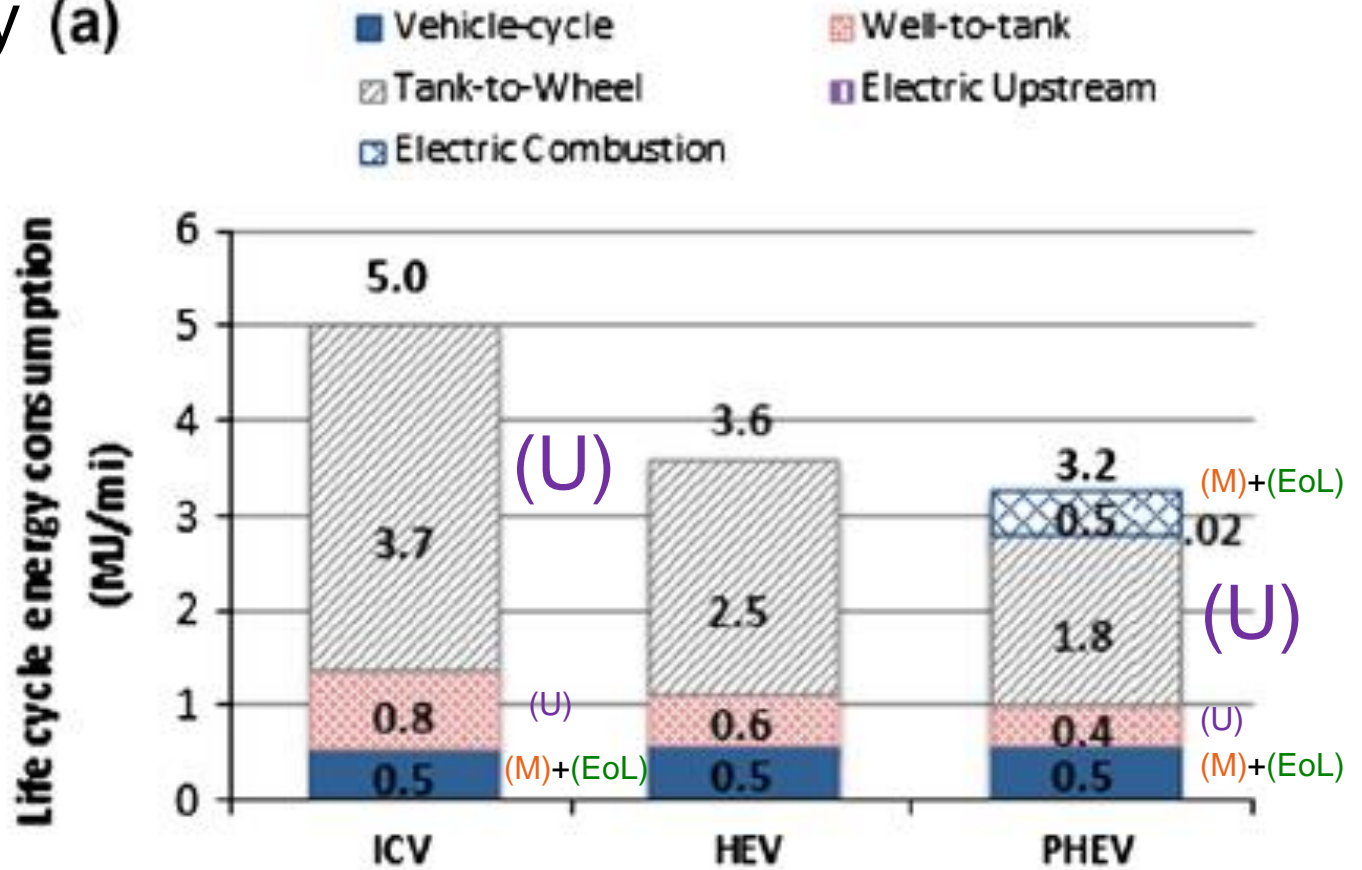
Flowchart source: Bras, B. "Sustainable Design and Manufacturing Including Environmental and Social Considerations," www.sdm.gatech.edu

Compiled by: R. Simmons, Strategic Energy Institute, Georgia Institute of Technology, 2021

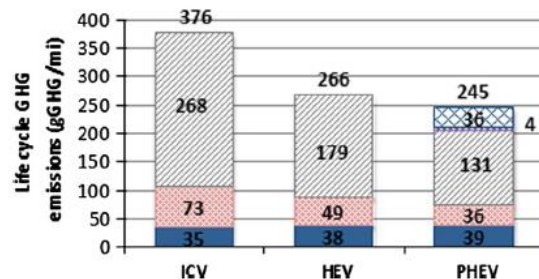
Energy/Emissions LCA for cars

LCA Energy (a)

(M) Mfg
(U) Use
(EoL) End of Life



LCA Emissions



Lewis, AM, Jarod CK, & Keoleian, GK. "Vehicle lightweighting vs. electrification: life cycle energy and GHG emissions results for diverse powertrain vehicles." *Applied Energy* 126 (2014): 13-20.

Surprising Fact

- ▶ 99.3% of Lead Acid Batteries are recycled in the US
- ▶ 5% of Lithium Ion Batteries are recycled in the US

Sources: DOE. 2019 and [wastedive.com](https://www.wastedive.com)

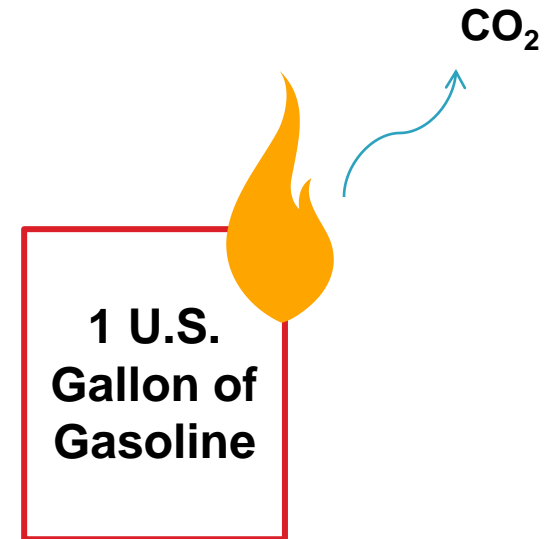
<https://www.energy.gov/sites/prod/files/2019/07/f64/112306-battery-recycling-brochure-June-2019%202-web150.pdf>

<https://www.wastedive.com/news/lithium-ion-battery-recycling-ev-li-cycle-retriev/608778/#:~:text=The%20industry%20is%20familiar%20with,recycled%2C%20according%20to%20the%20DOE.>

Poll Question 5

How much CO₂ is emitted from the combustion of 1 gallon of gasoline?

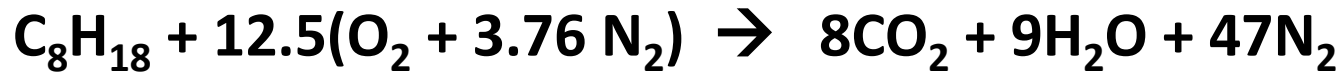
Note: 1 gallon of gasoline weighs about 6.3 lbs.



- A. Less than 3 lbs
- B. Between 3 and 7 lbs
- C. Between 7 and 16 lbs
- D. More than 16 lbs

Calculating the CO₂ released by 1 gallon of gasoline

Consider the combustion reaction of octane, a primary constituent of gasoline:



1 mol C₈H₁₈ ≈ 114 g/gmol

1 gallon ≈ 6.3 lbs = 2860 g

Thus, we have 2860/114 mols of fuel = 25.1 mols fuel

1 mol CO₂ ≈ 44 g/gmol

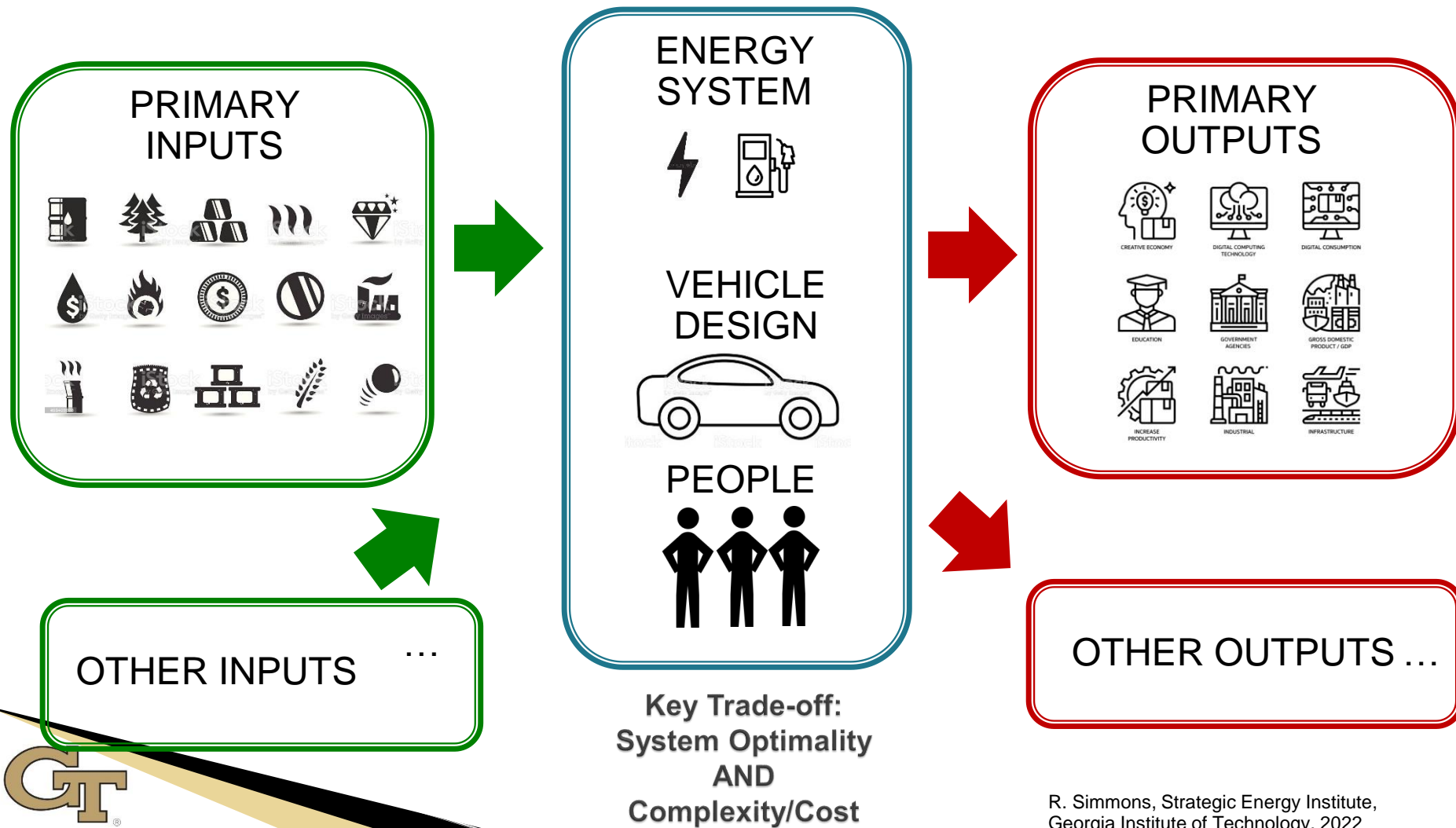
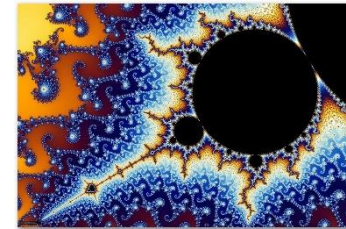
The reaction produces → 8 mols CO₂ for every mol of fuel

This equals 8*25.1 = 200.8 mols CO₂

→ 200.8 mols * 44g/gmol ≈ 8835 g ≈ **19.4 lbs**

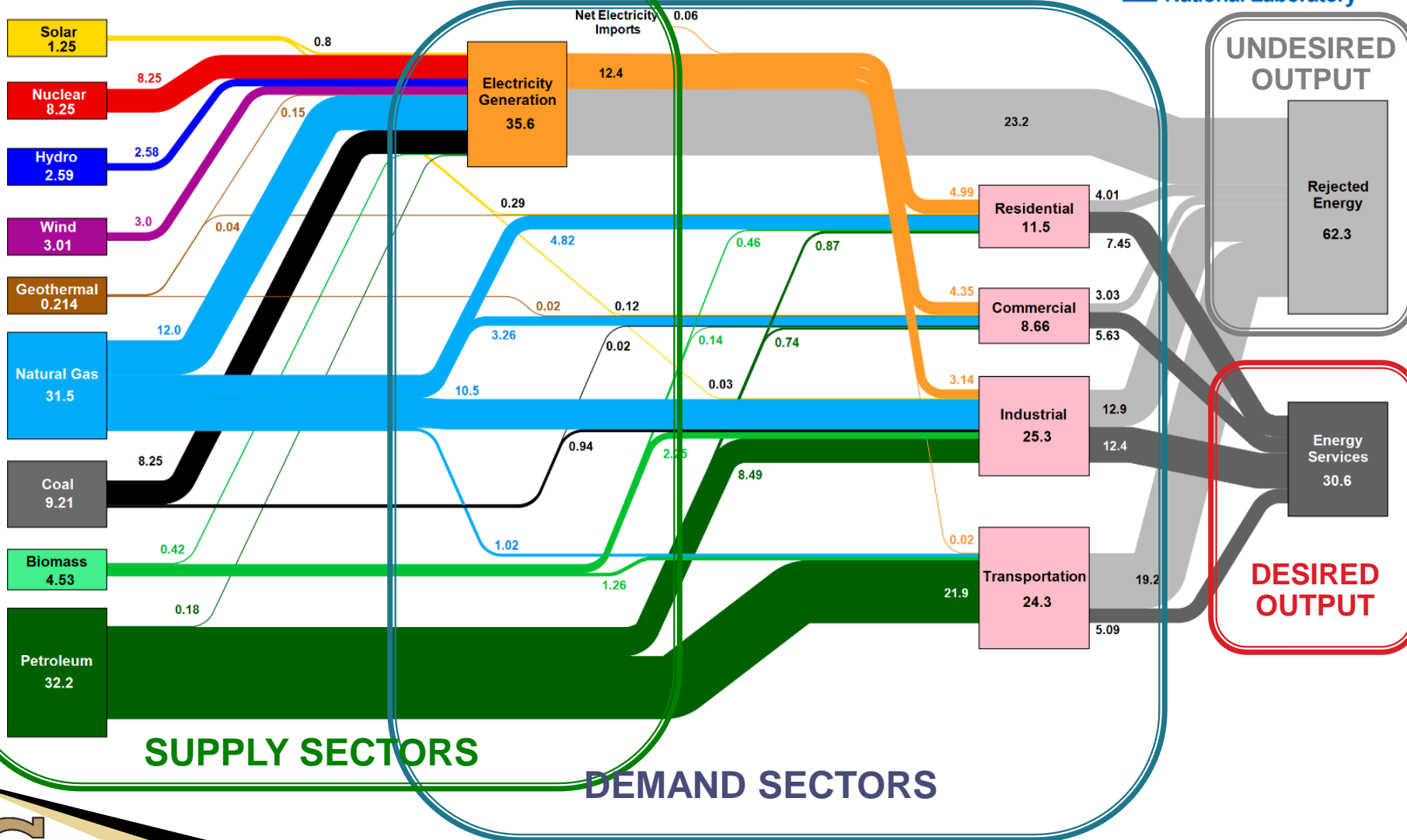
CO ₂ Tailpipe Emissions Fuel Type	CO ₂ Emissions Kg CO ₂ /Liter of Fuel	CO ₂ Emissions Lbs CO ₂ /gallon of fuel
Gasoline	2.3 kg/L	19.4 lbs/gal
Liquid Petroleum Gas (LPG)	1.6 kg/L	12.7 lbs/gal
Diesel	2.7 kg/L	22.2 lbs/gal

Engineering 101: System Boundary is Key



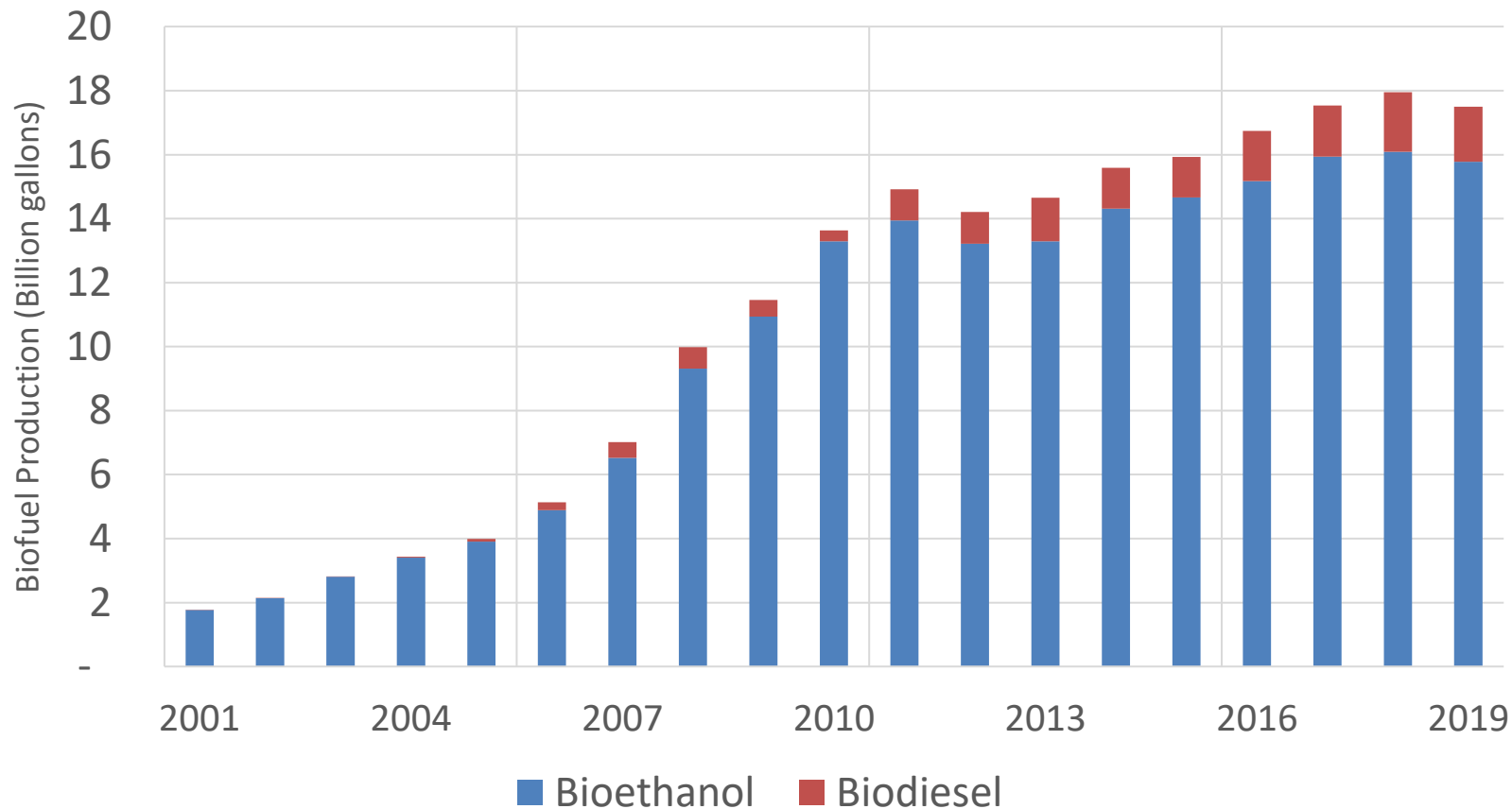
System Boundary: US Economy

Estimated U.S. Energy Consumption in 2020: 92.9 Quads



U.S. Biofuels Production

US Biofuels Production 2001-2019



Bioethanol: 15.8 BGPY
Biodiesel: 1.8 BGPY
All U.S. Ground Fuels: 215 BGPY
Share that is Biofuels: 8%

Source: USDA Economic Research Service, U.S. Bioenergy Statistics, 2021
<https://www.ers.usda.gov/data-products/us-bioenergy-statistics/>



Advanced Biofuels

Major 2nd Generation Biofuel Conversion Pathways:

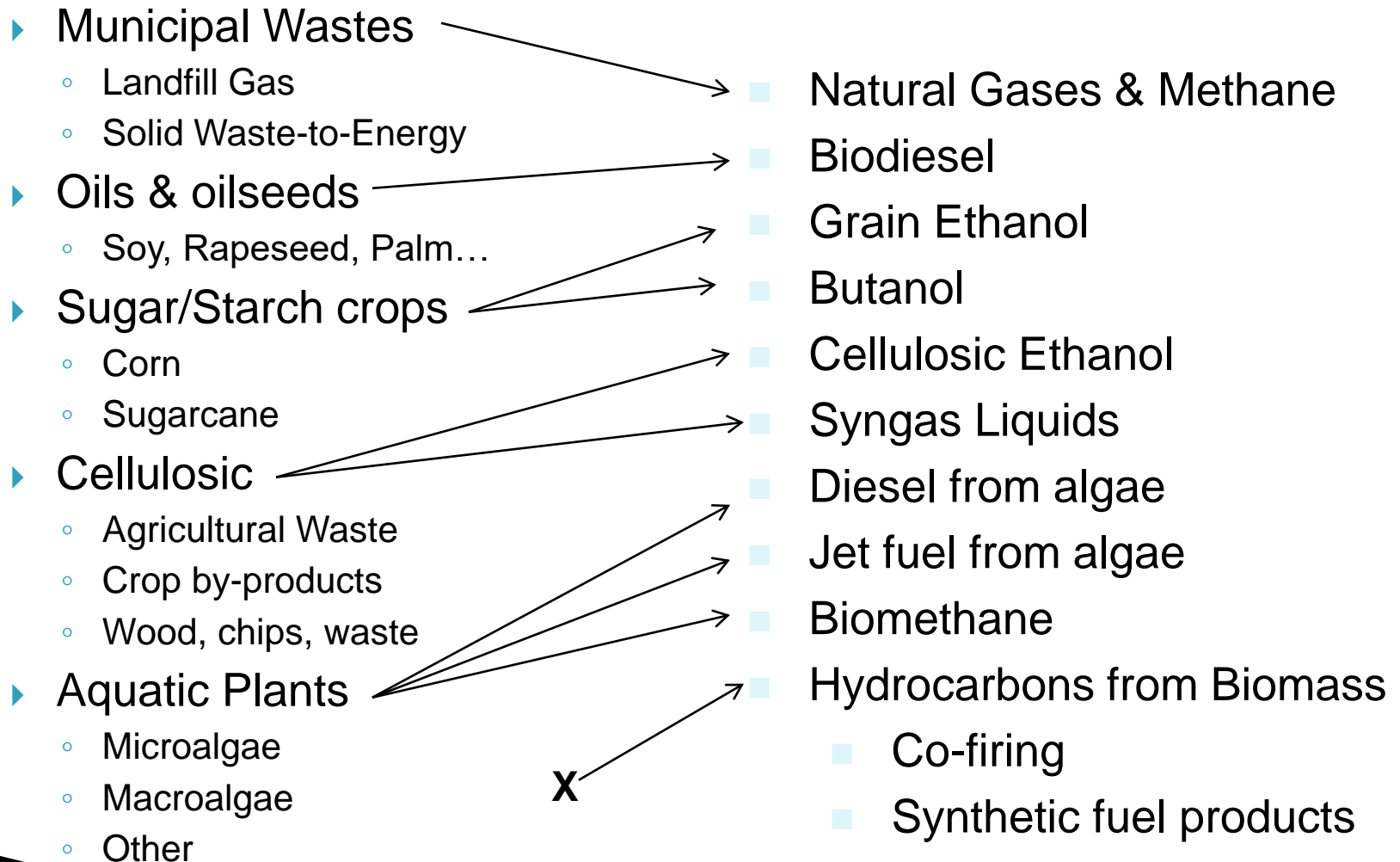
■ Biochemical

- Pre-treatment of biomass into cellulose, hemicellulose & lignin
- Enzymatic hydrolysis
- Biological conversion of sugars to alcohol via fermentation

■ Thermochemical

- Heating in the absence of air/oxygen
- Extraction of hydrocarbons into liquid (pyrolysis) or gas (syngas)
- Reformation, purification & production of synthetic hydrocarbons

Some 2nd Generation Biomass Feedstocks & Their Common Fuels

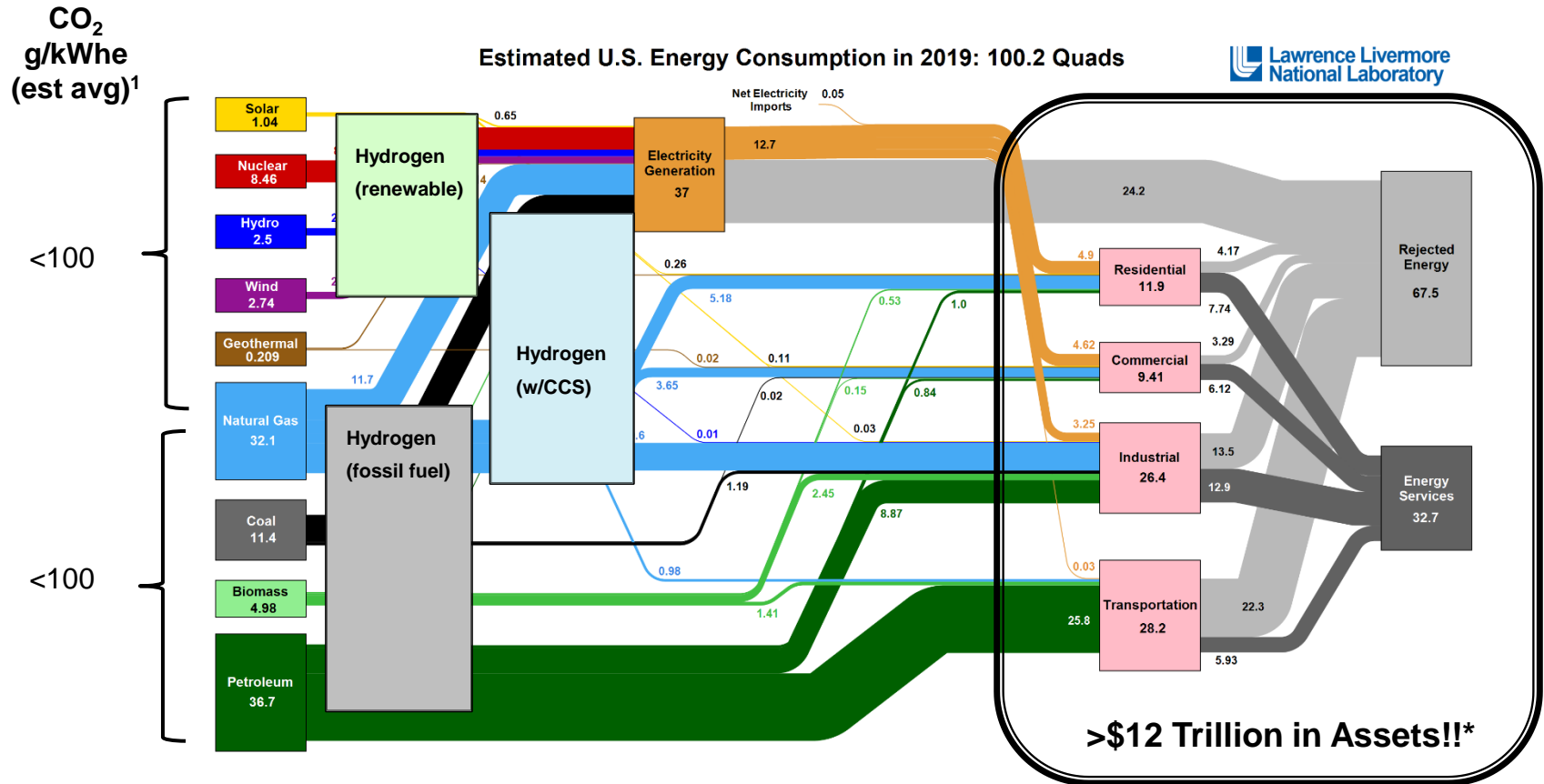


Common biofuels & their benefits...

Biofuel & Feedstock	Energy Balance	GHG reduction
Ethanol from Corn	1.4	20%
Ethanol from Sugarcane	8.0	60%
Cellulosic Ethanol	*6 - *14	*70- *90%
Biodiesel from Rapeseed	2.5	40%
Biodiesel from Soy	3.2	40%

Average Values from Published Literature,
*Cellulosic Ethanol Figures are estimates

Why is Hydrogen so enticing?



- ▶ Hydrogen can be an energy carrier or converted directly into electricity
- ▶ Hydrogen can be combusted directly in Gas Turbines
- ▶ Hydrogen can be combined with Carbon and synthesized into renewable fuels
- ▶ Hydrogen can help attain net zero CO₂ emissions

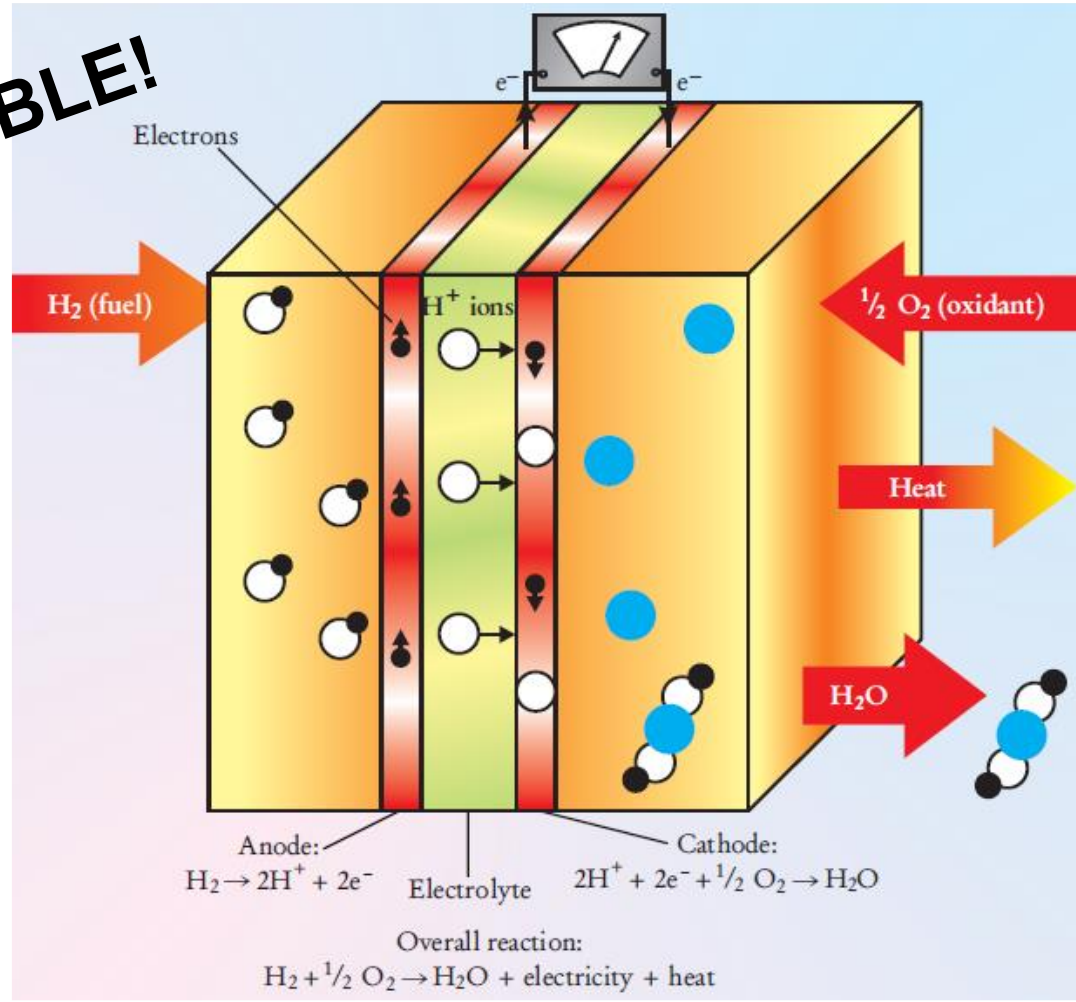


Source: Estimated Energy Use, US DOE, Lawrence Livermore National Lab, 2019 <https://flowcharts.llnl.gov/>

Source*: Simmons and Metz, Georgia Tech Strategic Energy Institute, 2019

Why is Hydrogen so enticing?

REVERSIBLE!



$\eta_{12} \approx 0.65$
(FC stack generating electricity)

$\eta_{21} \approx 0.90$
(Electrolysis of water to produce H_2)

- ▶ Fuel Cells (FCs) enable bi-directional conversion between hydrogen and electricity



H₂: So what's the catch?

▶ Technical Challenges

- Conversion
- Storage
- Safety
- Distribution
- Combustion characteristics
- Diffusion rate
- Fungibility (i.e., “drop-in”)
- ...

▶ Economic Challenges

- Capital cost
- Feedstock cost
- Storage cost & Leakage Loss
- Delivery cost
- Infrastructure implications
- ...

▶ Environmental Challenges

- Net Energy & CO₂ impact
- LifeCycle Assessment (LCA)
- Human Health
- Ecosystem Quality
- Resource Depletion
- Full Cost Accounting is complex

How Green are Electric Cars?

Comparing vehicles using dissimilar energy sources

**LARGELY
INDEPENDENT
OF LOCALITY**

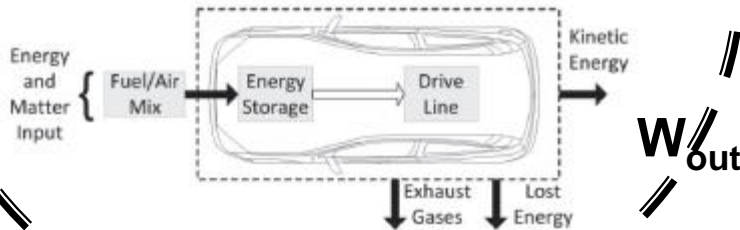
Again- Consider the
system boundary!
MPG \neq MPGe

**HIGHLY
DEPENDENT
ON
LOCALITY**



**INTERNAL
COMBUSTION
ENGINE VEHICLE**

Open Thermodynamic System

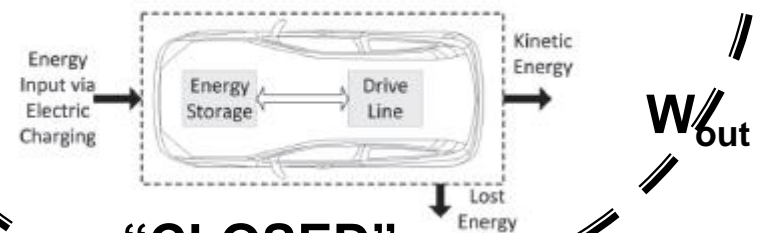


“OPEN”



**ELECTRIC
VEHICLE**

Closed Thermodynamic System



“CLOSED”



Comparison of Energy Use by Vehicle Type

VEHICLE TYPE

DRIVING CYCLE

TEMPERATURE

Modeling is based upon representative vehicles

- ▶ **Conventional**
- ▶ **ICE-SI** (Toyota Corolla, Honda Civic, Ford Focus)
- ▶ **ICE-CI** (Volkswagen Jetta)



- ▶ **Hybrid**
- ▶ **HEV** (Toyota Prius)



- ▶ **Plug-in hybrid**
- ▶ **PHEV40** (Chevrolet Volt)



- ▶ **Electric Vehicle**
- ▶ **EV** (Nissan Leaf)



Model Inputs

VEHICLE TYPE

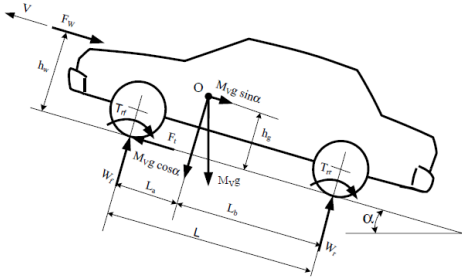
DRIVING CYCLE

TEMPERATURE

Specifications	Vehicle Type:	ICE-SI ⁷	ICE-CI ⁸	HEV-PS ⁹	PHEV-40 ¹⁰	EV-PAC ¹¹
	Source:	[8 ^{a,b,c}]	[8 ^d]	[8 ^a]	[8 ^e]	[8 ^f]
Vehicle Attribute						
Vehicle mass ¹ [kg]		1438	1595	1519	1857	1610
Drag coefficient		0.29	0.30	0.25	0.29	0.28
Frontal area [m ²]		2.12	2.10	2.17	2.16	2.31
Engine power ² [kW]		108	104	73	63	-
Electric motor power ² [kW]		-	-	60	111	80
Total vehicle power² [kW]		108	104	100	111	80
Battery mass ³ [kg]		-	-	45	198	294
Battery capacity ³ [kWh]		-	-	1.3	16.5	24.0
Fuel economy ⁴ [US ⁵ mpg]		31.4	34.0	50.0	37.0	-
Fuel consumption ⁴ [L/100km]		7.5	6.9	4.7	6.4	-
Elec. consumption ⁶ [Wh/km]		-	-	-	214	184
Equiv. fuel econ. ⁶ [mpge]		-	-	-	98	114
All electric range [km(mi)]		-	-	-	64(40)	134(84)



Propulsion model



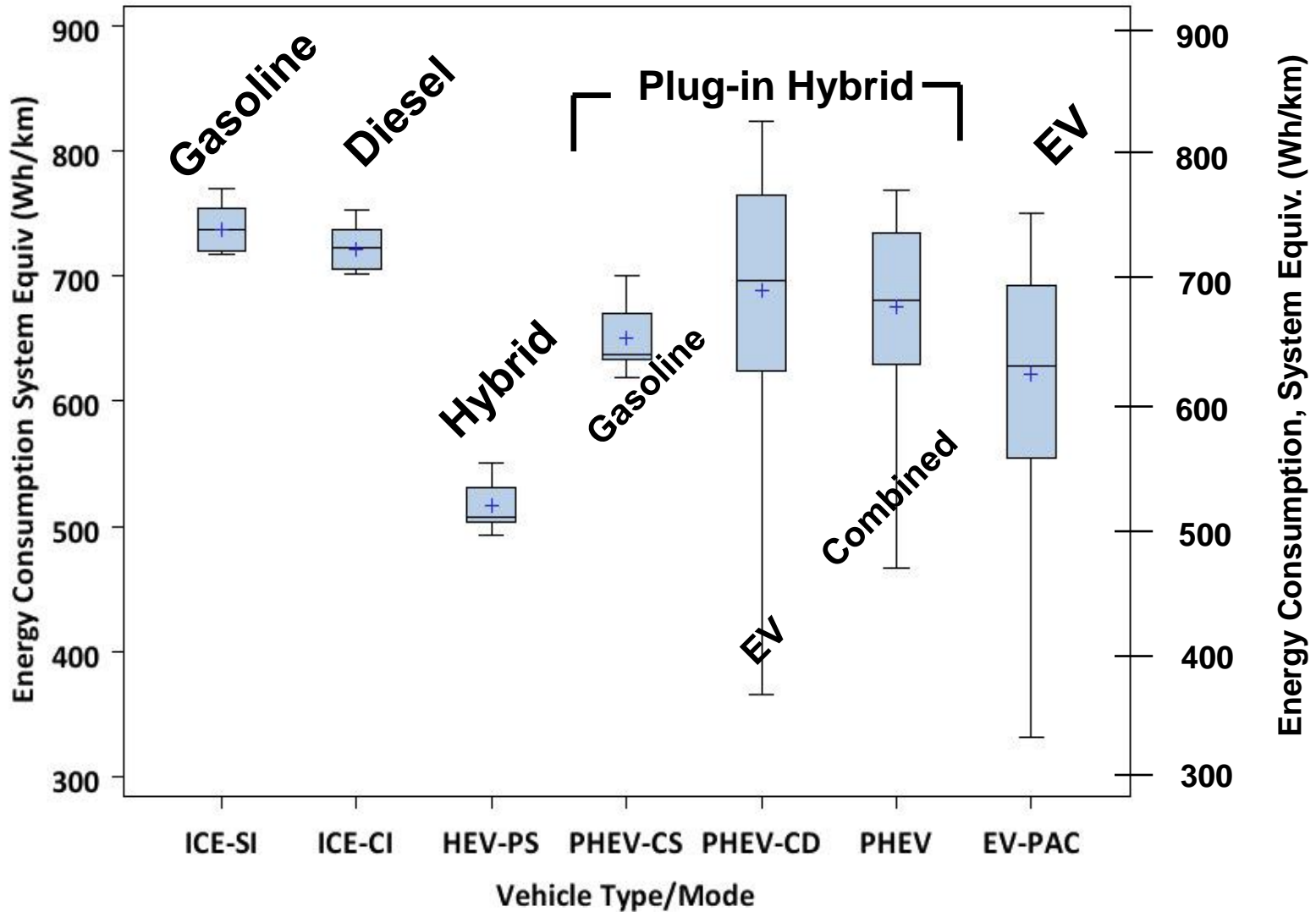
General dynamic force balance:

$$F_{tractive} = F_{rolling_resistance} + F_{aero_drag} + F_{hill_climb} + F_{acceleration} \quad \text{Eq. (9)}$$

$$F_{tractive} = m_{veh} \cdot g(C_0 + C_1 v_{veh}) + 0.5 \rho_{air} C_{Drag} A_F v_{veh}^2 + m_{veh} g \cdot \sin \theta_{grade} + m_{veh} \cdot a_{veh} \quad \text{Eq. (10)}$$

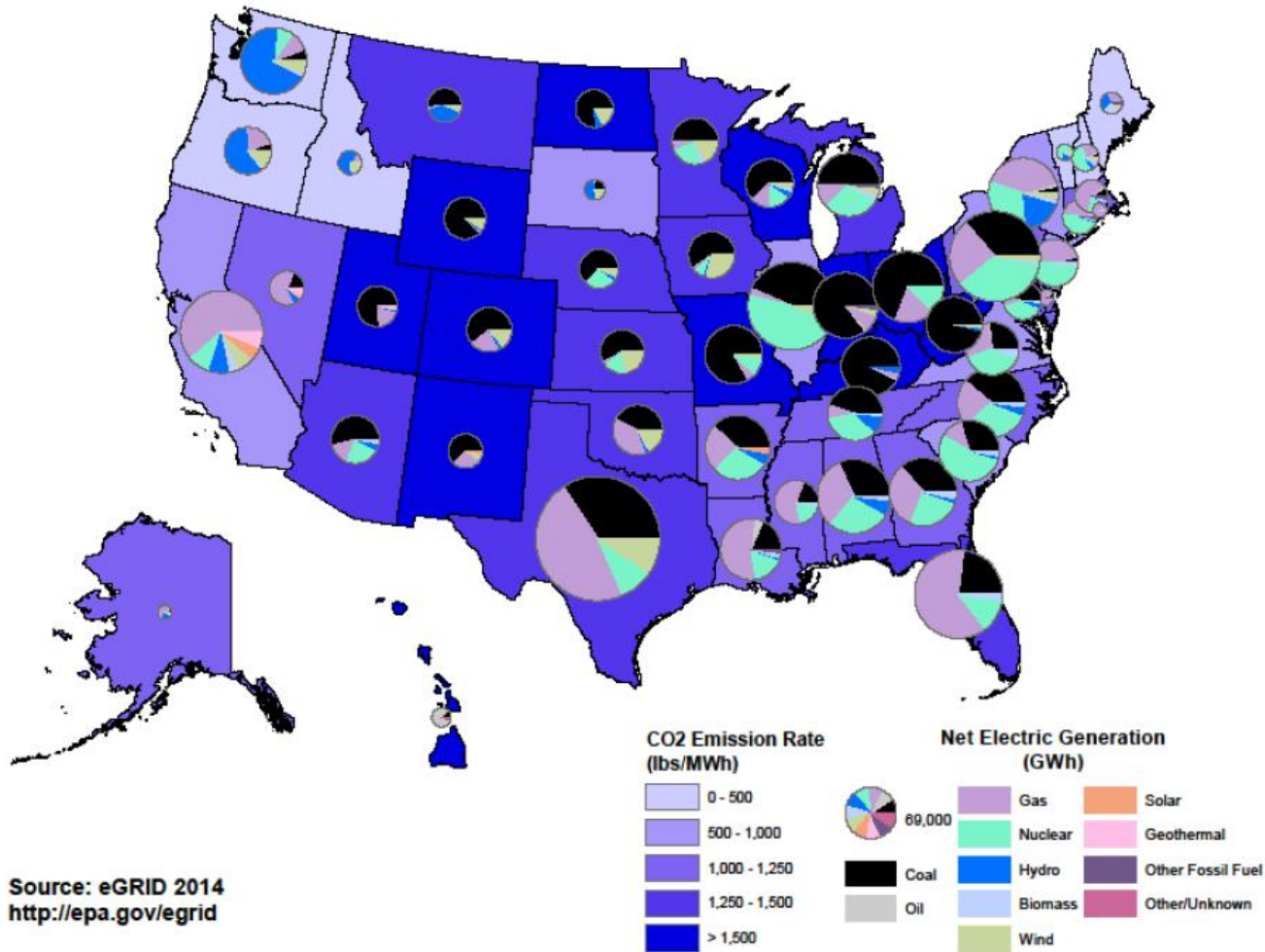
$$P_{tractive} = F_{tractive} \cdot v_{veh}(t) \quad \text{Eq. (11)}$$

Energy use depends on vehicle type and locality



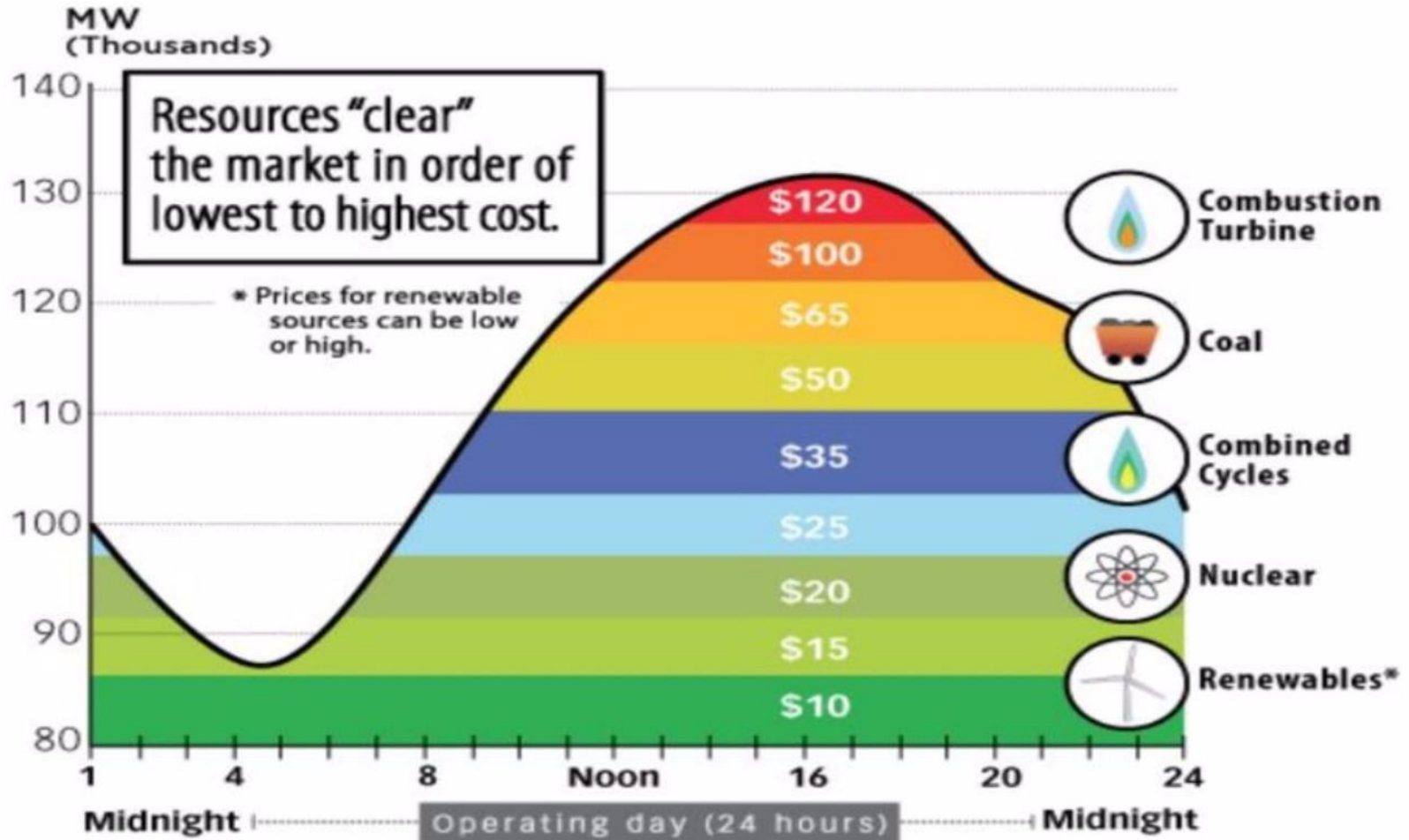
State-by-state generation by source 2014

12. Generation by Fuel Type and CO₂ Emission Rates (eGRID2014v2)



(illustrative)

Example Grid Dispatch Curve

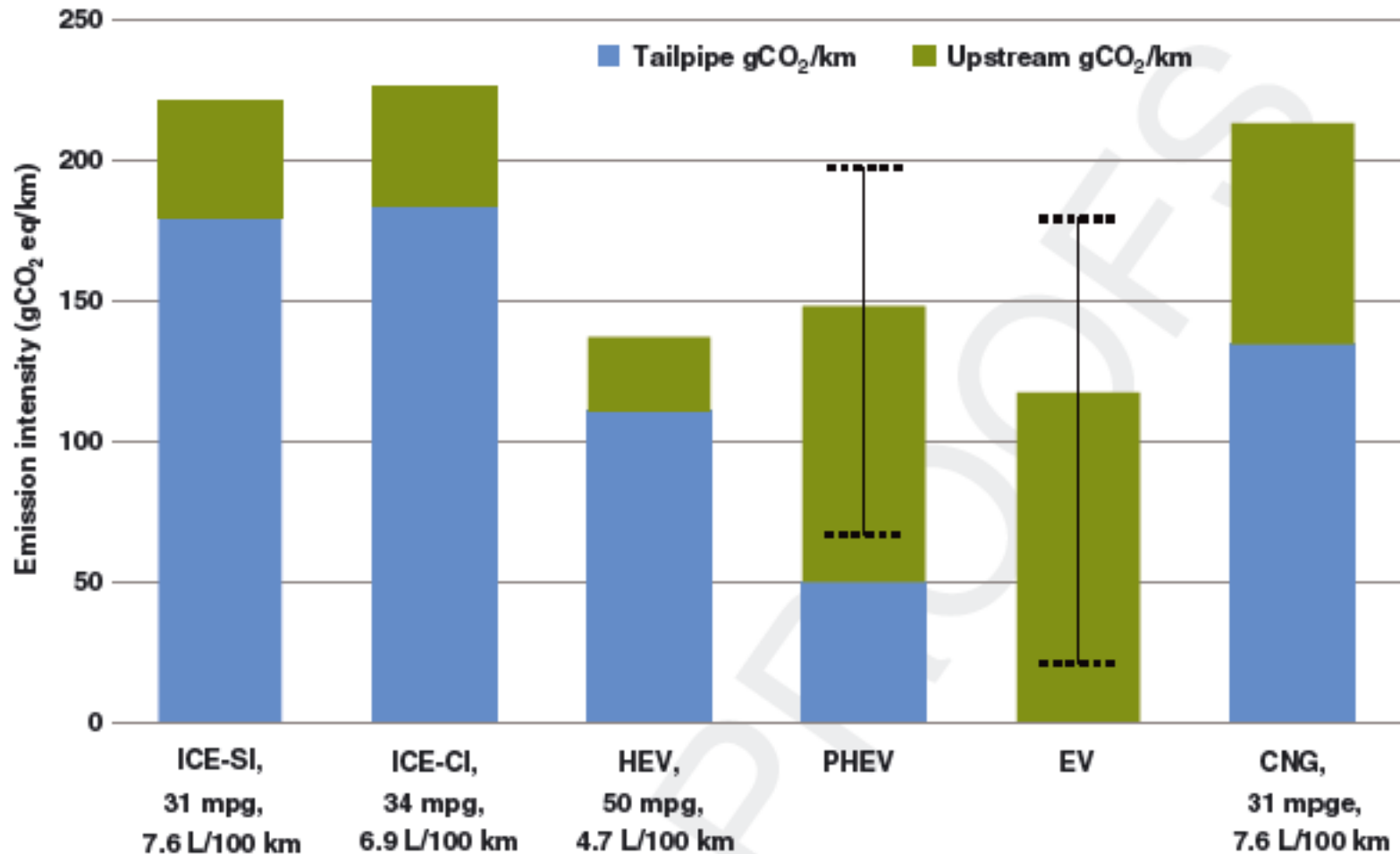


Important Note: This is MARGINAL cost per unit (i.e., doesn't consider capex), Not Levelized Cost, or LCOE



Graphic Source: Northbridge Energy Partners, P. Kelly-Detwiler

Use phase emissions depend on: vehicle type and locality



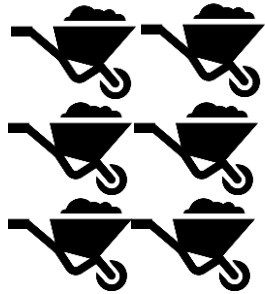
Wide variation for grid-recharged EVs

Upstream energy impact

Resource

On-board

Delivered Work



6.0



5.0

▶ Gasoline Car



1.0



4.0



3.2

▶ Hybrid Car



1.0



6.0

Coal (worst case)



2.5

Renewables (best case)



1.7

▶ Electric Car

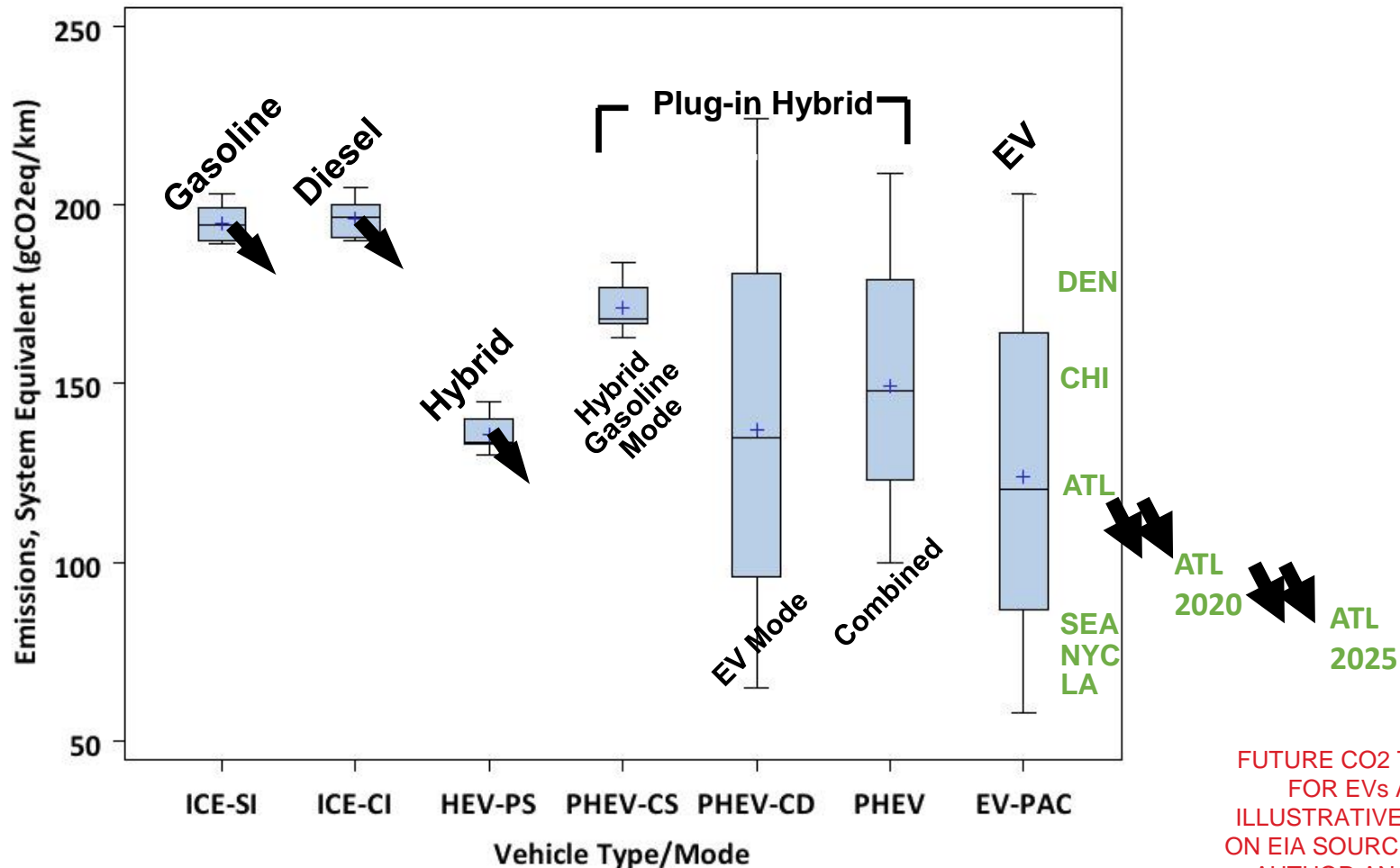


1.0

Caveat: Treats all energy resources as interchangeable



System-Equivalent CO₂ Emissions by Vehicle Type and Locality



FUTURE CO₂ TRENDS FOR EVs ARE ILLUSTRATIVE, BASED ON EIA SOURCE DATA & AUTHOR ANALYSIS

- ▶ Total CO_{2eq} emissions from extracted resource thru use-phase consumption
- ▶ Does NOT include CO₂ associated with vehicle manufacturing

51

Notes: Grid data as of 2014.

Simmons, Richard A. "A techno-economic investigation of advanced vehicle technologies and their impacts on fuel economy, emissions, and the future fleet." PhD diss., Purdue University, 2015.

Additional Info, As of 2020: <https://cepl.gatech.edu/projects/Drawdown-Georgia>

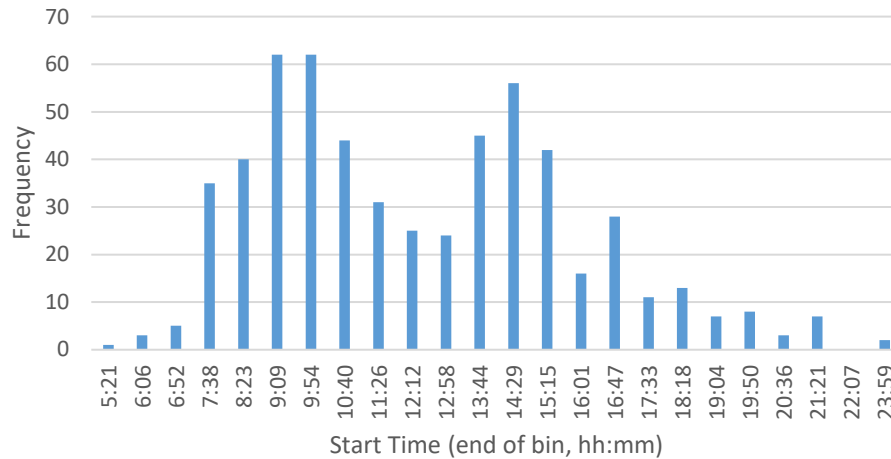


How Green are Electric Cars?

▶ EV Charging fn(time)

- Sample Data
- Georgia Tech Daytime Workplace Charging

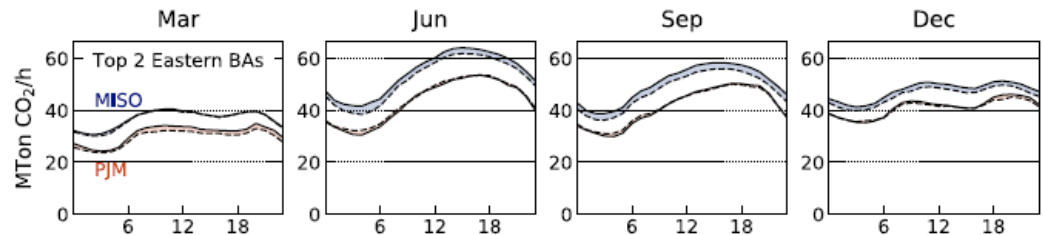
Charging Start Time Histogram, Feb 2020 GT



R. Simmons, Strategic Energy Institute, Georgia Institute of Technology, 2022

▶ Emissions fn(time,location)

- Top 2 Eastern Balancing Authorities
- Typical Marginal CO₂ emissions
- Seasonal and Hourly



Tracking emissions in the US electricity system

Jacques A. de Chalendar, John Taggart, Sally M. Benson

Proceedings of the National Academy of Sciences Dec 2019, 116 (51) 25497-25502; DOI: 10.1073/pnas.1912950116

Resources of Interest

LCA examples

▶ e-Scooters

- Hollingsworth, Joseph, Brenna Copeland, and Jeremiah X. Johnson. "Are e-scooters polluters? The environmental impacts of shared dockless electric scooters." *Environmental Research Letters* 14, no. 8 (2019): 084031.

▶ Reusable vs. Disposable Cups

- Hocking, M. Reusable and disposable cups: An energy-based evaluation. *Environmental Management* (1994) 18(6):889-899.
- Woods, L, Bakshi, B. Reusable vs. disposable cups revisited: guidance in life cycle comparisons addressing scenario, model, and parameter uncertainties for the US consumer. *Int J Life Cycle Assess* (2014) 19:931–940.

▶ Videoconferences vs. In-person meetings

- Ong, Dennis, Tim Moors, and Vijay Sivaraman. "Comparison of the energy, carbon and time costs of videoconferencing and in-person meetings." *Computer communications* 50 (2014): 86-94.

▶ Locality impacts of EV efficiency and emissions

- Tamayao, Mili-Ann M., Jeremy J. Michalek, Chris Hendrickson, and Inês ML Azevedo. "Regional variability and uncertainty of electric vehicle life cycle CO2 emissions across the United States." *Environmental science & technology* 49, no. 14 (2015): 8844-8855.

Resources of potential interest

<https://epicenter.energy.gatech.edu/energy-101/>

- ▶ Energy 101 SE
 - Open Access 8-part Video Short Course



Resources of Interest

► Hydrogen Readings

Applied Energy 281 (2021) 115958

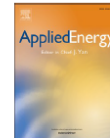


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journal homepage: www.elsevier.com/locate/apenergy



Uncovering the true cost of hydrogen production routes using life cycle monetisation

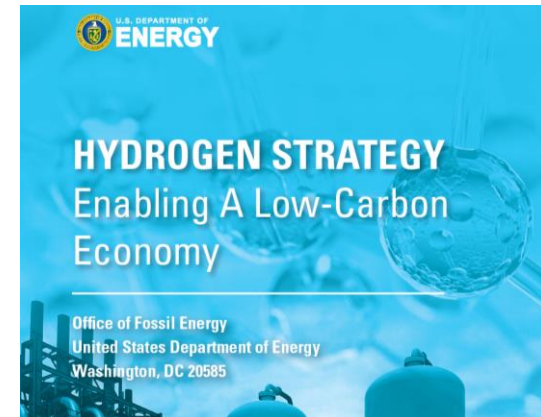
Amjad Al-Qahtani^a, Brett Parkinson^a, Klaus Hellgardt^a, Nilay Shah^a, Gonzalo Guillen-Gosalbez^{b,*}

^a Centre for Process Systems Engineering, Department of Chemical Engineering, Imperial College London, South Kensington Campus, London SW7 2AZ, UK

^b Institute for Chemical and Bioengineering, Department of Chemistry and Applied Biosciences, ETH Zürich, Vladimir-Prelog-Weg 1, 8093 Zürich, Switzerland

Al-Qahtani, Amjad, et al. "Uncovering the true cost of hydrogen production routes using life cycle monetisation." *Applied Energy* 281 (2021): 115958.

Blog on similar boiling water experiments and comparisons
<http://insideenergy.org/2016/02/23/boiling-water-ieq/>



US. DOE, Office of Fossil Energy "Hydrogen Strategy: Enabling a Low-Carbon Economy," Washington DC. June 2020

Resources of potential interest

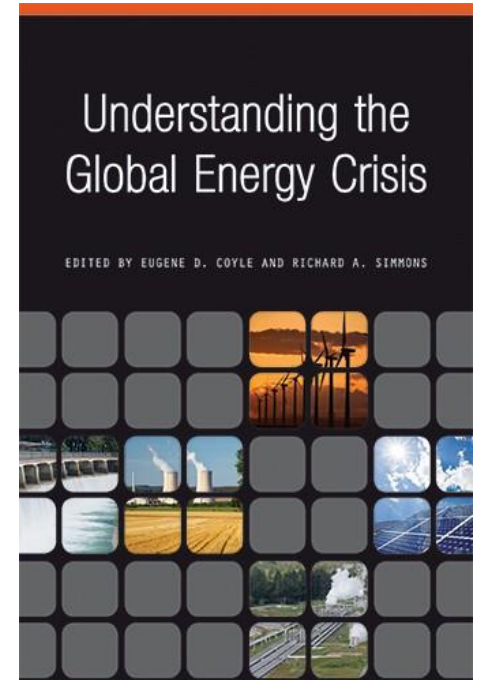
“Understanding the Global Energy Crisis”

Edited by Coyle & Simmons

Published by Purdue University Press, 2014

Available for free download:

http://docs.lib.purdue.edu/purduepress_ebooks/29/



Thank you

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